

**HPV Data Set**

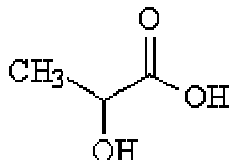
**LACTIC ACID**

**CAS # 50-21-5**

**Dossier number 50215**

## Substance information

<b>CAS No.</b>	50-21-57
<b>EINECS Name:</b>	lactic acid
<b>EC No.</b>	200-018-0
<b>TSCA Name:</b>	Propanoic acid, 2-hydroxy-, (2S)-
<b>Molecular Formula</b>	C <sub>3</sub> H <sub>6</sub> O <sub>3</sub>



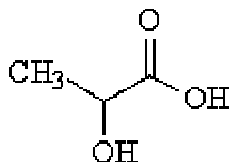
<b>IUPAC Name:</b>	LACTIC ACID
<b>Mol. Weight:</b>	90
<b>Physical status:</b>	liquid or white crystals
<b>Colour:</b>	colourless to slightly yellow
<b>Odour:</b>	nearly odourless

**Most of the lactic acid nowadays is supplied in the L(+) form, the natural form.**

**PURAC only manufactures the L(+) form.**

**Substance information for the L(+) lactic acid is as follows:**

<b>CAS No.</b>	79-33-4
<b>EINECS Name:</b>	L(+) lactic acid
<b>EC No.</b>	201-196-2
<b>TSCA Name:</b>	Propanoic acid, 2-hydroxy-, (2S)-
<b>Molecular Formula</b>	C <sub>3</sub> H <sub>6</sub> O <sub>3</sub>



<b>IUPAC Name:</b>	L(+) LACTIC ACID
<b>Mol. Weight:</b>	90
<b>Physical status:</b>	liquid or white crystals
<b>Colour:</b>	colourless to slightly yellow
<b>Odour:</b>	nearly odourless

**This HPV data file typically contains information about L(+) lactic acid.**

# Chapter 1                      Physico-chemical Data

## ▪ **Melting Point**

Value:                                      ≤ 54 degree C  
Decomposition:                          no at < 110 degree C  
Sublimation:                            no  
Method:                                    other  
GLP:                                        no data  
Test substance:                        100% crystalline pure product used

## ▪ **Melting Point of the anhydrous form of the racemic mixture**

Kraft and Dyes found for the racemic lactic acid the melting point 18°C., Borsook, Huffman and Liu reported 16.8°C and more recently Lockwood, Yoder and Zienty reported values of up to 28-33°C. When one of the optically active isomers is present in excess, the melting point is higher; this indicates that racemic lactic acid forms a eutectic mixture.

Ref: ***Lactic Acid: properties and chemistry of lactic acid and derivatives by C.H. Holten, 1971, page 21***

## ▪ **Boiling Point / Vapour pressure**

Boiling point:                            ca. 258 degree C at 1000 hPa  
Vapour pressure:                        ca. .0041 hPa at 20 degree C  
Decomposition:                        yes  
Method:                                    other  
GLP:                                        no data  
Remark:                                    solutions can polymerise on boiling

It is not possible to determine this value very accurate, because lactic acid tends to polymerise by a polycondensation reaction.

The boiling point of lactic acid at difference vapour pressures according to

- Aspen Database
- PURAC In-house Pure compound database

**Table** Boiling points of lactic acid at different pressures (Aspen Database)

Pressure (mbar)	Boiling temperature (°C)
1	79.6
5	101.9

10	112.8
50	141.8
100	156.2
200	172.1
400	189.8
600	201.0
800	209.4
1013	216.6

The vapour pressures of lactic acid as a function of the boiling temperature according to the Aspen formula (PLXANT).

$$PLXANT = e^{C1 + \frac{C2}{C3+T} + C4 \cdot T + C5 \cdot \ln T + C6 \cdot T^{C7}}$$

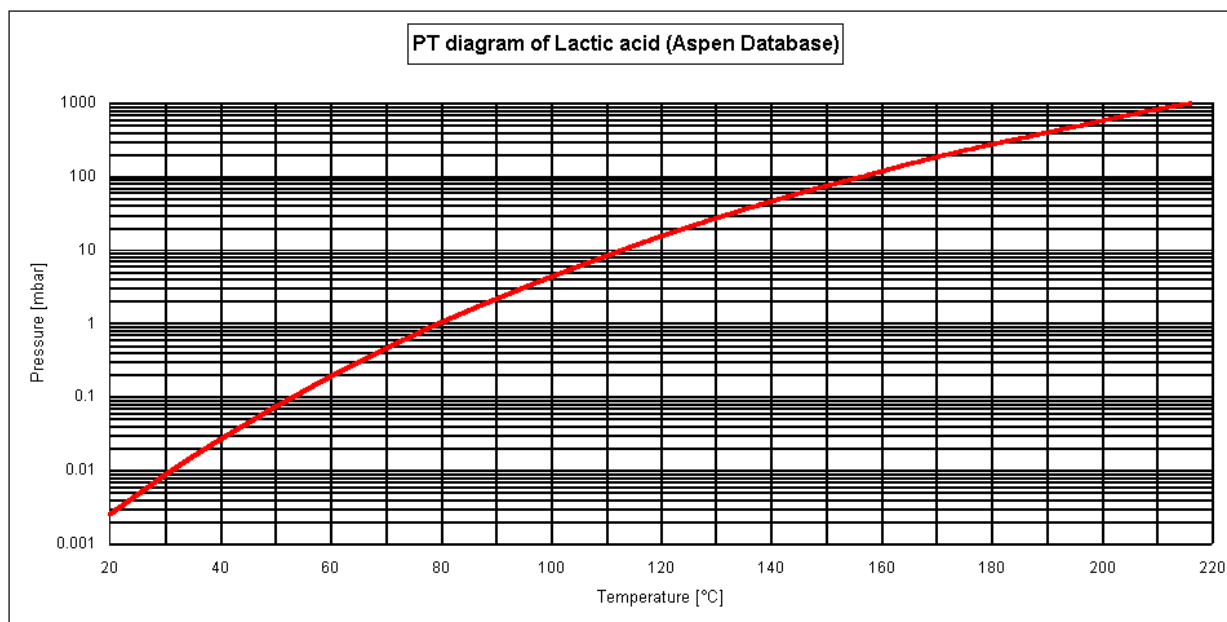
: Vapour pressure (N/m<sup>2</sup> or Pa)

PLXANT

C1, C2 C3, C4, C5, C6, C7 : Regression coefficients for chemical compound

T : Boiling temperature (K)

**Figure** Vapour Pressure diagram



The regression coefficients are:

C1 = 225.19

C5 = -28.816

C2 = -18757

C6 = 0.000012998

C3 = 0

C7 = 2

C4 = 0

G.P.v.Lieshout measured the vapour pressure of lactic acid between 0 and 180 °C.

**Table** Boiling points of lactic acid at different pressures

Temperature (°C)	Vapour Pressure (mbar)
0	0.0005
10	0.0015
20	0.0041
30	0.0107
40	0.0258
50	0.0588
60	0.1273
70	0.2629
80	0.5198
90	0.9877
100	1.8093
110	3.2042
120	5.5004
130	9.1736
140	14.8959
150	23.5938
160	36.5163
170	55.3114
180	82.1111

The vapour pressures of lactic acid as a function of the boiling temperature were fitted with the Aspen formula (PLXANT). The estimated atmospheric boiling point is about 258 °C.

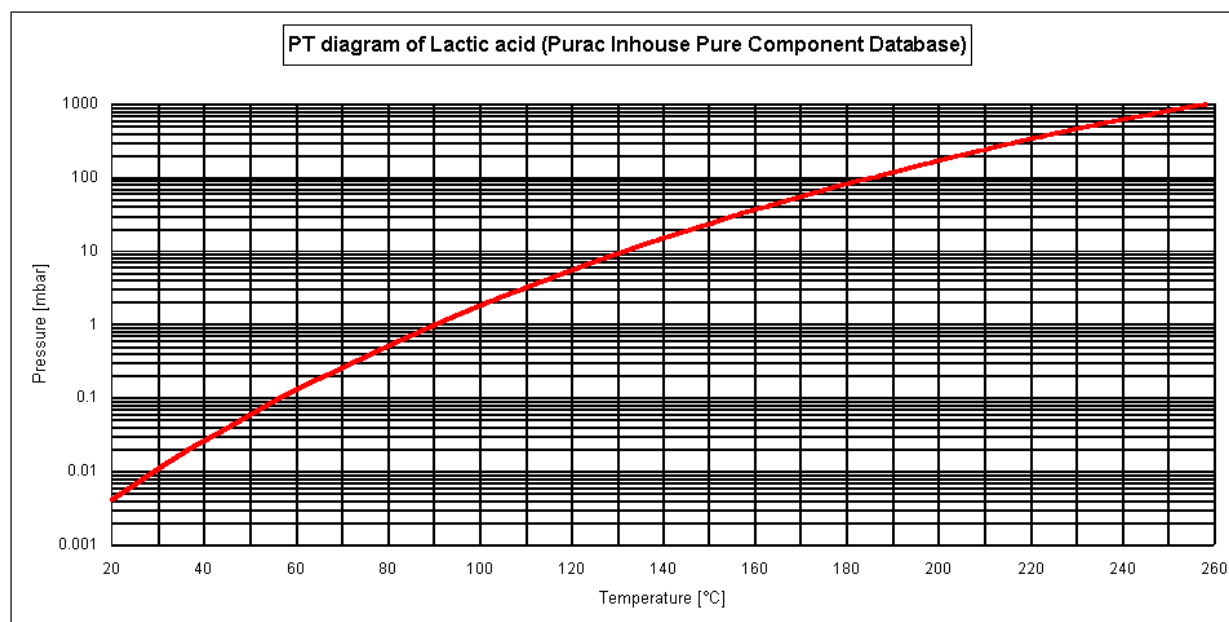
$$\text{PLXANT} = e^{\frac{C1 + \frac{C2}{C3+T}}{C4 \cdot T + C5 \cdot \ln T + C6 \cdot T^2}}$$

: Vapour pressure (N/m<sup>2</sup> or Pa)

PLXANT

C1, C2 C3, C4, C5, C6, C7

: Regression coefficients for chemical compound



: Boiling temperature (K)

Figure: Vapour Pressure diagram

The regression coefficients are:

C1 = 32.4649                      C5 = -0.376037

C2 = -8835.5                    C6 = 0

C3 = 0                            C7 = 0

C4 = -0.003654

## ▪ Density

Type: density  
 Value:  $\geq 1.2255 \text{ g/cm}^3$  at 20 degree C  
 Method: other  
 GLP: yes  
 Test substance: Lactic acid 80%

Holten describes the liquid density of lactic acid solutions in water.

**Table** Liquid Densities of aqueous lactic acid solutions at different temperatures.

Concentration (wt%)	Temperature (°C)										
	0	10	20	30	40	50	60	70	80	90	100
0	1.0004	0.9997	0.9979	0.9953	0.9919	0.9878	0.9831	0.9778	0.9720	0.9658	0.9591
10	1.0269	1.0251	1.0223	1.0188	1.0145	1.0097	1.0042	0.9983	0.9919	0.9851	0.9779
20	1.0535	1.0505	1.0468	1.0424	1.0373	1.0317	1.0256	1.0190	1.0120	1.0047	0.9970
30	1.0802	1.0762	1.0715	1.0662	1.0604	1.0540	1.0472	1.0400	1.0325	1.0246	1.0164
40	1.1069	1.1019	1.0964	1.0902	1.0836	1.0766	1.0692	1.0614	1.0533	1.0449	1.0363
50	1.1333	1.1275	1.1211	1.1142	1.1069	1.0992	1.0912	1.0829	1.0743	1.0655	1.0564
60	1.1591	1.1524	1.1453	1.1377	1.1299	1.1216	1.1131	1.1043	1.0953	1.0861	1.0766
70	1.1835	1.1762	1.1685	1.1604	1.1521	1.1434	1.1345	1.1254	1.1160	1.1065	1.0967
80	1.2061	1.1983	1.1902	1.1818	1.1731	1.1642	1.1550	1.1456	1.1360	1.1263	1.1163
90	1.2263	1.2183	1.2100	1.2014	1.1925	1.1834	1.1741	1.1646	1.1550	1.1451	1.1351
95	1.2354	1.2274	1.2190	1.2104	1.2015	1.1924	1.1831	1.1736	1.1639	1.1541	1.1441
100	1.2438	1.2357	1.2274	1.2188	1.2099	1.2008	1.1916	1.1822	1.1725	1.1627	1.1527

## Liquid Density of pure lactic acid

The liquid density of pure lactic acid at difference temperatures according to:

- Aspen Database
- PURAC In-house Pure Component Database

**Table** Densities of lactic acid at different temperatures (ASPEN Database)

Temperature (°C)	Density (g/ml)
20	1.2255
30	1.2163
40	1.2069
50	1.1974
60	1.1878
70	1.1780
80	1.1681
90	1.1580
100	1.1478
110	1.1374
120	1.1268
130	1.1160
140	1.1051
150	1.0939
160	1.0825

The liquid densities of lactic acid as a function of the temperature according to the Aspen equation for liquid density (DNLDIP):

$$\text{DNLDIP} = \frac{C1}{C2^{1+(1-\frac{T}{C3})^{C4}}}$$

DNLDIP : Liquid density (kmole/m<sup>3</sup>)

C1, C2 C4 : Regression coefficients for chemical compound

C3 : Critical temperature (K)

T : Temperature (K)

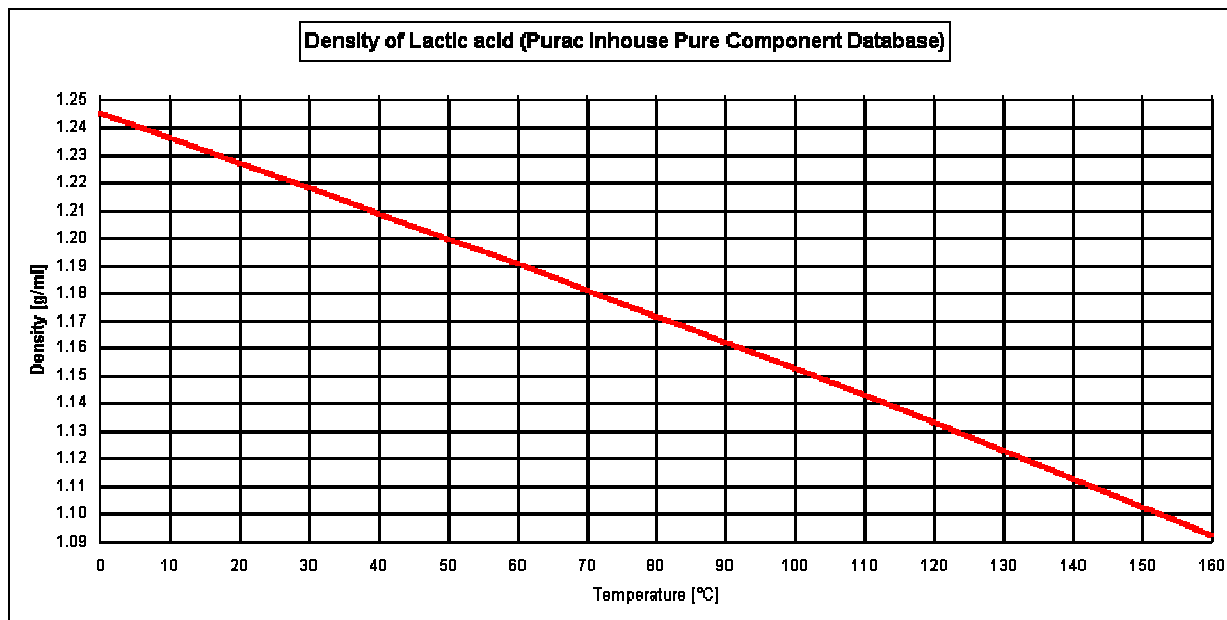
The regression coefficients are:

C1 = 0.9798                      C3 = 675

C2 = 0.24593                    C4 = 0.2333

### Liquid density diagram





**Table** Densities of lactic acid at different temperatures (Purac In-house Pure Component Database)

Temperature (°C)	Density (g/ml)
0	1.2453
10	1.2364
20	1.2274
30	1.2183
40	1.2092
50	1.2000
60	1.1907
70	1.1813
80	1.1719
90	1.1623
100	1.1527
110	1.1429
120	1.1330
130	1.1231
140	1.1130
150	1.1028
160	1.0924

The liquid densities of lactic acid as a function of the temperature according to the Aspen equation for liquid density (DNLDIP):

$$\text{DNLDIP} = \frac{C1}{C2^{1 + (1 - \frac{T}{C3})^{C4}}}$$

: Liquid density (kmole/m<sup>3</sup>)

DNLDIP

C1, C2 C4 : Regression coefficients for chemical compound

C3 : Critical temperature (K)

T : Temperature (K)

The regression coefficients are:

C1 = 1.61248 C3 = 729.9

C2 = 0.314479 C4 = 0.328317

#### ▪ **Partition Coefficient**

Partition Coeff.: octanol-water

log Pow: ca. -.62 at 20 degree C

Method: OECD Guide-line 117 "Partition Coefficient (n-octanol/water),  
HPLC Method"

Year: 1987

GLP: yes

#### ▪ **Solubility in water**

Value: ca. 100 vol% at 25 degree C

pH value: ca. 1.2

pKa: 3.68 at 25 degree C

Descr.: of very high solubility

Method: other

GLP: no data

Deg. product: not measured

Stable: yes

Remark: completely soluble at 25 degrees C

J.v.Krieken determined the phase diagram of lactic acid/ water. Due to practical problems the diagram wasn't completed, but the missing part was extrapolated.

**Table** Solubility of monomeric lactic acid in water

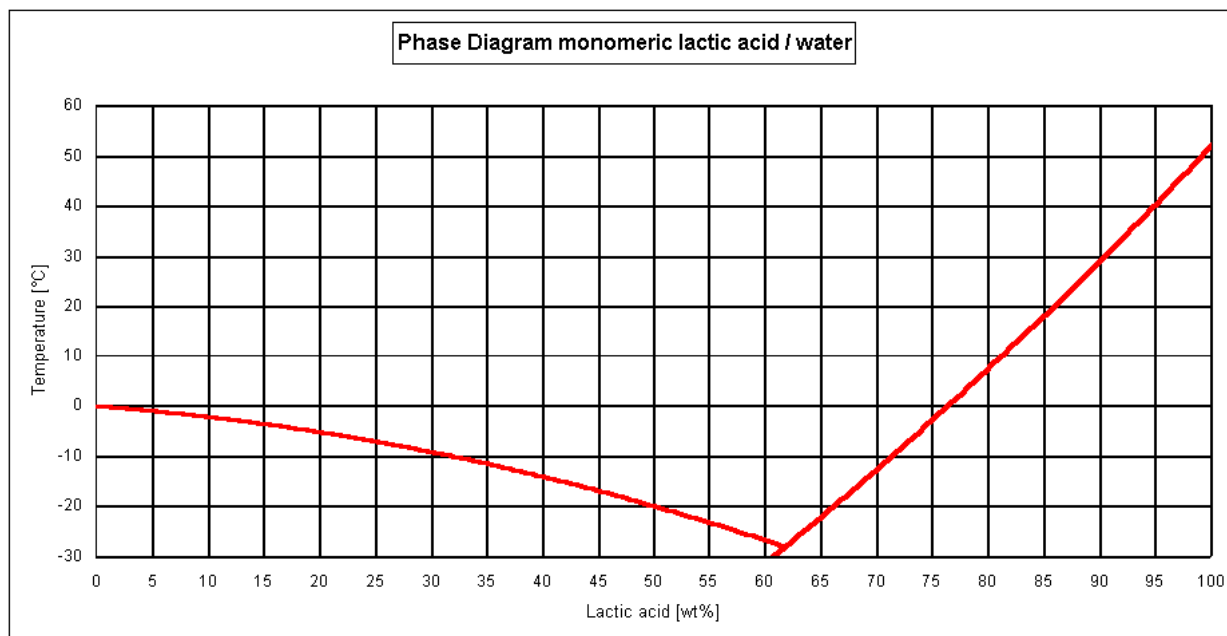
Temperature (°C)	Lactic acid (wt%)
-20	65.9
-15	68.8
-10	71.5
-5	74.2
0	76.7
5	78.6
10	81.3
15	83.5
20	86.1
25	87.6
35	92.7
40	95.1

**Table** Freezing point data of monomeric lactic acid in water

Temperature (°C)	Lactic acid (wt%)
------------------	-------------------

0	0
-5	18.6
-10	32.5
-15	41.9
-21	51.7

The eutectic point of lactic acid/water was calculated by extrapolation and is ca. 61.7 wt% lactic acid and -27.1 °C.



**Figure** Phase Diagram

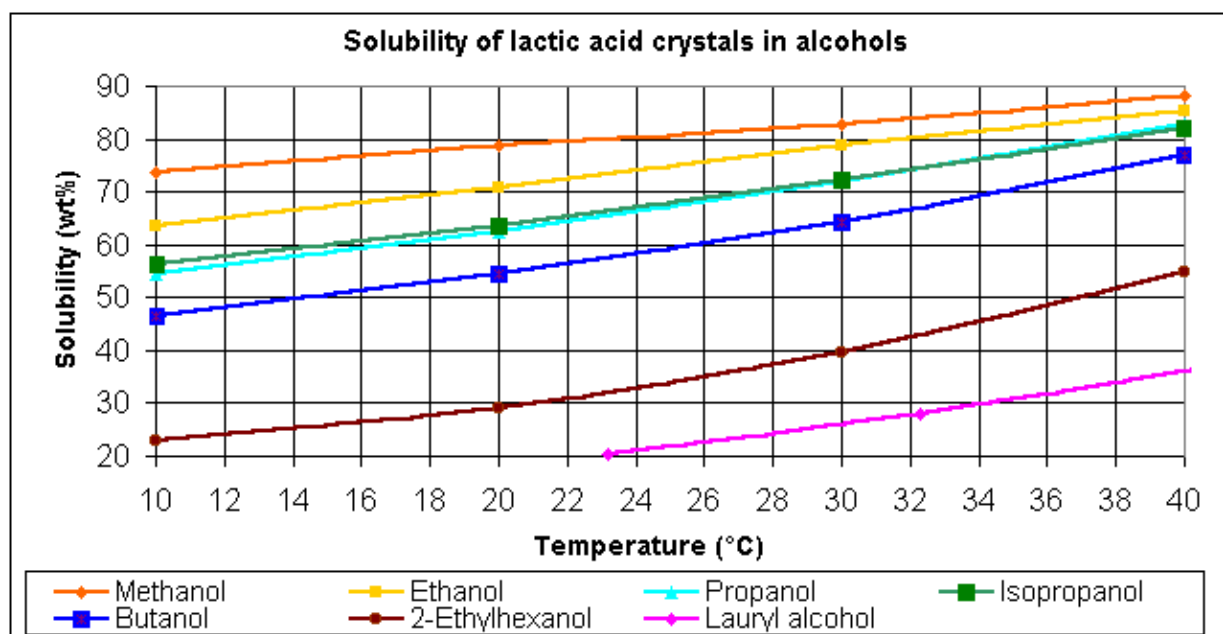
### Solubility of lactic acid in other solvents

J.v. Krieken investigated the solubility of pure lactic acid in a range of solvents by stirring the specific solvent with an excess of (S)-lactic acid crystals at a specific temperature. After separation of the solids from the liquid, the clear liquid was analysed on free acidity.

**Table** solubility of lactic acid in a range of solvents

Solvent	Solubility (wt%) of solution			
	10 °C	20 °C	30 °C	40 °C
Methanol	73.8	78.6	82.8	88.1
Ethanol	63.6	70.9	78.7	85.2
1-Propanol	54.5	62.4	71.7	82.7
2-Propanol	56.1	63.4	72.2	82.2
1-Butanol	46.3	54.5	64.3	77.1
2-Ethyl-1-hexanol	22.9	29.0	39.6	54.9
Cyclohexane				0.04
Hexane	< 0.01	< 0.01	< 0.01	0.02

Toluene	0.06	0.11	0.24	0.50
Ethyl lactate	37.0	45.9	57.2	72.8
Butyl lactate	27.7	35.7	47.3	64.8
2-Ethylhexyl lactate	15.5	20.9	30.9	46.7
Ethyl acetate	27.1	39.9	56.2	72.9
Diethyl ether	23.8	38.7	59.8	
Diisopropyl ether	6.4	9.2	15.9	44.4
Tetrahydrofuran	58.4	65.1	72.4	82.9
Dichloromethane	0.59	1.02	2.7	
Chloroform	0.31	0.67	1.67	59.7
2-Butanone	40.7	52.9	67.3	81.6
Acetone	53.4	61.4	71.5	82.9



## ▪ Viscosity

Test type: other: ASPEN database ; PURAC internal  
Value: ca. 53.52 mPa s (dynamic) at 20 degree C

The liquid viscosity of lactic acid at difference temperatures according to

- Aspen Database
- PURAC In-house Pure Component Database

The liquid viscosity of aqueous lactic acid solutions in water

- 0 - 88 wt% of Lactic Acid
- 90 - 110 wt% Lactic Acid

**Table** Viscosity of lactic acid at different temperatures (Aspen Database)

Temperature (°C)	Viscosity (cP)
20	53.52
30	33.26
40	21.29
50	14.01
60	9.44
70	6.51
80	4.59
90	3.29
100	2.40
110	1.78
120	1.34
130	1.03
140	0.79
150	0.62
160	0.49

The liquid viscosity of lactic acid as a function of the temperature according to the Aspen equation for liquid viscosity (MULDIP):

$$\text{MULDIP} = e^{C1 + \frac{C2}{T} + C3 \ln T + C4 \cdot T^{C5}}$$

: Liquid viscosity (N\*s/m<sup>2</sup> or Pa.s)

MULDIP

C1, C2 C3, C4, C5 : Regression coefficients for chemical compound

T : Temperature (K)

The regression coefficients are:

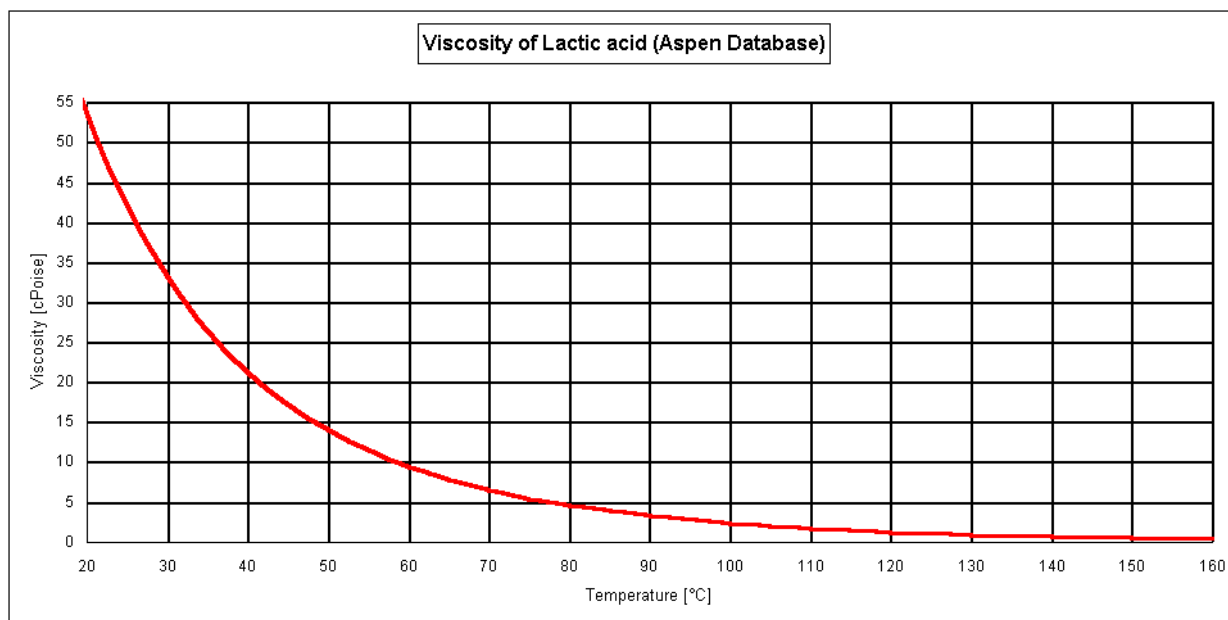
C1 = -14.403

C2 = 4097.9

C3 = -0.4407

C4 = 0

C5 = 0



**Figure:** Liquid Viscosity diagram

Table 12 Viscosity of lactic acid at different temperatures (Purac In-house Pure Component Database) <sup>2</sup>

Temperature (°C)	Viscosity (cP)
20	1342.33
30	579.14
40	280.59
50	150.73
60	88.80
70	56.81
80	39.14
90	28.83
100	22.54

The liquid viscosity of lactic acid as a function of the temperature according to the Aspen equation for liquid viscosity (MULDIP):

MULDIP:                    Liquid viscosity ( $\text{N}\cdot\text{s}/\text{m}^2$  or  $\text{Pa}\cdot\text{s}$ )

C1, C2, C3, C4, C5: Regression coefficients for chemical compound

T:                            Temperature (K)

The regression coefficients are:

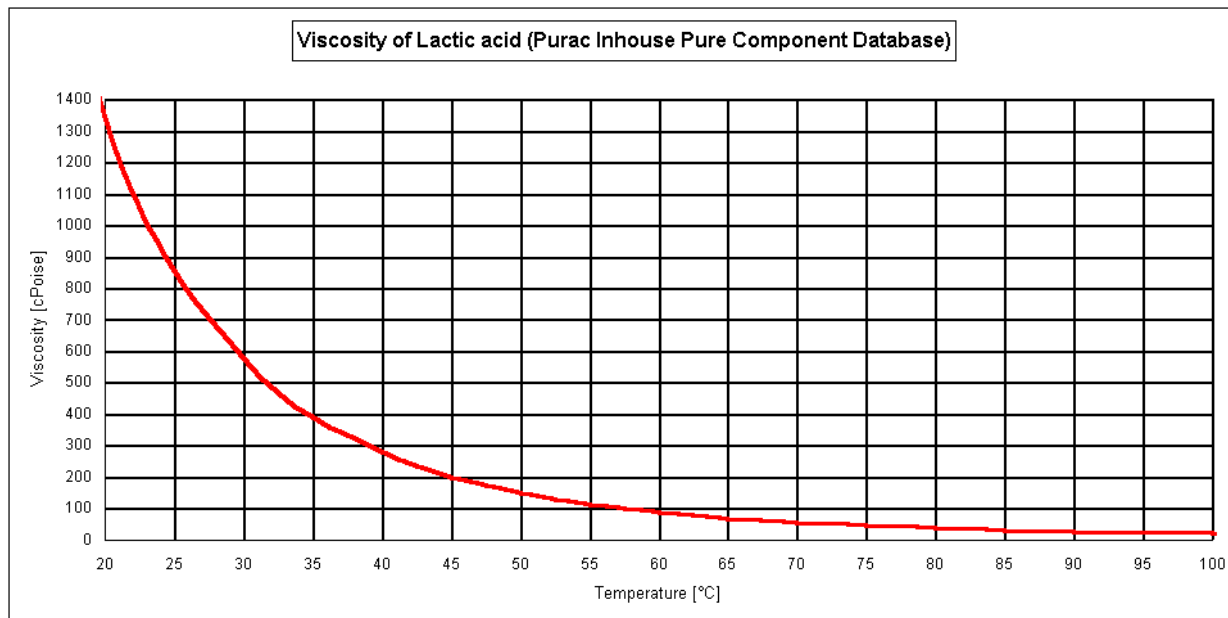
C1:                        421.094

C2:                        25091.4

C3:                        59.1119

C4:                        0

C5:                        0



**Figure:** Liquid Viscosity diagram

## ***References Chapter 1: Physico chemical data***

- National Chemical Inventories  
Codex NCINF5  
ISSN: 1089-6279
- Handbook of Chemistry and Physics  
WEAST  
66th Edition
- ASPEN database
- PURAC internal databank

## **Chapter 2      Environmental Fate and Pathways**

### ▪ **Photodegradation**

The photochemical oxidisation of lactic acid is discussed in "Lactic acid properties and chemistry of lactic acid and derivatives by C.H. Holten (1971)". The first observation that lactic acid is photosensitive was made in 1910 by Berthelot and Gaudechon, who irradiated calcium lactate and ethyl lactate with ultraviolet rays. They observed decomposition with the formation of gas containing carbon monoxide, carbon dioxide, hydrogen and methane.

Recognized method, i.e. OECD: Modelling conducted, no guideline studies used.

Method: Estimated Programs Interface (EPIWIN V3.05, Atmospheric Oxidation Program v 1.90). Model executed in October 2002.

Results / observations: The AOP component of EPIWIN was used to calculate the rate of photodegradation for L(+) lactic acid.

SMILES : O=C(O)C(O)C

CHEM : Propanoic acid, 2-hydroxy-, (S)-

MOL FOR: C3 H6 O3

MOL WT : 90.08

----- SUMMARY (AOP v1.90): HYDROXYL RADICALS -----

Hydrogen Abstraction = 5.2598 E-12 cm<sup>3</sup>/molecule-sec

Reaction with N, S and -OH = 0.6600 E-12 cm<sup>3</sup>/molecule-sec

Addition to Triple Bonds = 0.0000 E-12 cm<sup>3</sup>/molecule-sec

Addition to Olefinic Bonds = 0.0000 E-12 cm<sup>3</sup>/molecule-sec

Addition to Aromatic Rings = 0.0000 E-12 cm<sup>3</sup>/molecule-sec

Addition to Fused Rings = 0.0000 E-12 cm<sup>3</sup>/molecule-sec

OVERALL OH Rate Constant = 5.9198 E-12 cm<sup>3</sup>/molecule-sec

HALF-LIFE = 1.807 Days (12-hr day; 1.5E6 OH/cm<sup>3</sup>)

HALF-LIFE = 21.682 Hrs

----- SUMMARY (AOP v1.90): OZONE REACTION -----

\*\*\*\*\* NO OZONE REACTION ESTIMATION \*\*\*\*\*

(ONLY Olefins and Acetylenes are Estimated)

Experimental Database: NO Structure Matches

### ▪ **Stability in Water**

Lactic Acid (88% and 60% aqueous solutions) were investigated.

"The kinetics of degradation of Lactic Acid was done at elevated temperature, since the decomposition rates of lactic acid, (..) were too slow to obtain kinetic data within



reasonable time. At the condition studied (25, 40, 80 and 120 degrees centigrade) the decompositions of these compounds followed apparent first order kinetics because the mean correlation coefficient was above 0.980.

Lactic acid was very stable in aqueous solutions at 80 degrees centigrade (less than 30% decomposition after 175 at 80 degrees) and degradation was not different when combined with the various excipients tested. The shelf lives determined for lactic acid ranged from 79 years when combined with isopropyl palmitate to 98 years when combined with sorbic acid.

#### ▪ **Transport between Environmental Compartments**

Type: other: see free texts RM.

Remark: Lactic acid is not volatile and it has a high biodegradation rate. Therefor transport between compartments is no issue for this compound.

No experimental data are available on fugacity. However, calculation using the sotware program EPI Suite version 3.1 gave the following information which supports the statement given above:

Level III Fugacity Model:			
	Mass Amount (%)	Half-Life (hr)	Emissions (kg/hr)
Air	3.16	43.4	1000
Water	46.3	208	1000
Soil	50.5	208	1000
Sediment	0.0691	832	0
Persistence Time: 222 hr			

#### • **Biodegradation**

Value 50% degradation after 5 days and 67% after 20 days.

Breakdown Product: It is to be expected that L(+) lactic acid will be taken up into the metabolism of the bacteria in the activated sludge. Degradation products will therefor be CO<sub>2</sub> and H<sub>2</sub>O.

Method: BOD (Biochemical Oxygen Demand) and COD (Chemical Oxygen Demand) determinations were carried out for L(+) lactic acid usig the method described in teh Dutch guidenlines "water determination of biochemical oxgen demand after n days (BODn)" (NEN 6634) and "Water determination of chemical oxygen demand (COD)" (NEN 6633) respectively, these methods are similar to those referred in the EC Test Guidelines C.8 and C.9 Two concentrations (2 mg/L and 4 mg/L) were tested. An oculum was prepared from activated sludge. Its microbial activity appeared to be sufficient although the control substance glucose and glutamic acid had a BOD5 of slightly less than the required value of 4.00 ± 0.75 mg O<sub>2</sub>/L

## ***References Chapter 2: Environmental fate and pathways***

- Lactic Acid: properties and chemistry of lactic acid and derivatives by C.H. Holten, 1971, page 38
- TNO report R 92/018: BOD and COd of L(+) lactic acid according to EC test guidelines C.8 and C.9
- The ecotoxicity and the biodegradability of lactic acid, alkyl esters and lactate salts  
C.T. Bowmer et.al.  
Chemosphere, volume 37, No 7, pp 1317 - 1333, 1998
- Stability of lactic acid and glycolic acid in aqueous systems subjected to acid hydrolysis and thermal decomposition.  
M.M. de Villiers et.al.  
Journal of the society of cosmetic chemists, 48, 165-174 (August 1998)
- Handbook of Chemistry and Physics  
WEAST  
66th Edition

## **Chapter 3      Ecotoxicity**

### ▪ ***Acute/Prolonged Toxicity to Fish***

- Species: Brachydanio rerio (2.1 ±0.17 cm and 0.07 ±0.018 g)
- Report number: TNO R 91/295
- Type: Semistatic
- Exposure period: 48 hour(s)
- Test substance: 80% L(+) Lactic Acid
- Nominal concentrations: 0, 100, 180, 320, 560 and 1000 mg/l
- Analytical monitoring: yes
- 48h LC50: = 320 - calculated
- Limit Test: no
- Method: OECD Guideline 203 "Fish, Acute Toxicity Test"
- Water temperature: Between 24.0 and 24.7 °C
- Water hardness: About 220 mg/l expressed as CaCO<sub>3</sub>.
- Alkalinity: Unknown
- Total Organic content: 2.0 mg/l
- Dissolved oxygen: > 7.3 mg/l
- pH levels: During test between 3.2 and 8.2
- Year: 1992
- GLP: yes
- Remark: test solutions are not neutralised. It is more than likely that the low pH value affected the survival of the fishes.

- Species: Brachydanio rerio (2.1 ±0.17 cm and 0.07 ±0.018 g)
- Report number: TNO R 91/295
- Type: Semistatic
- Exposure period: 96 hour(s)
- Test substance: 80% L(+) Lactic Acid
- Nominal concentrations: 0, 100, 180, 320, 560 and 1000 mg/l
- Analytical monitoring: yes
- 96h NOEC: = 320 mg/l - measured/nominal
- 96h LC50: = 320 mg/l - calculated
- 96h LC100: = 560 mg/l - calculated
- Method: OECD Guideline 203 "Fish, Acute Toxicity Test"
- Water temperature: Between 24.0 and 24.7 °C
- Water hardness: About 220 mg/l expressed as CaCO<sub>3</sub>.
- Alkalinity: Unknown
- Total Organic content: 2.0 mg/l
- Dissolved oxygen: > 7.3 mg/l
- pH levels: During test between 3.2 and 8.1
- Year: 1992
- GLP: yes
- Remark: test solutions are not neutralised. It is more than likely that the low pH value affected the survival of the fishes.

- Species: Lepomis macrochirus (24± 2.3 mm and 0.37 ± 0.15 g)  
Report number: Analytical Bio Chemistry lab # 32146  
Type: Static  
Exposure period: 96 hour(s)  
Test substance: 80% L(+) Lactic Acid  
Nominal concentrations: 0, 56, 100, 180, 320 and 560 mg/l  
Analytical monitoring: yes  
NOEC: = 56 mg/l - measured/nominal  
24 h LC50: = 140 mg/l - measured/nominal  
48 h LC50: = 130 mg/l - measured/nominal  
96h LC50: = 130 mg/l - measured/nominal  
Limit Test: yes

Method: :

Ten fish were exposed to each test concentration and control.

The procedures for static bioassay described in (1) and (2) below were used in this experiment.

(1) Committee on methods for Toxicity Tests with Aquatic Organisms (C.E. Stephan chairman). 1975.

Methods for acute toxicity tests with fish, macro invertebrates and amphibians.

Environmental Protection Agency, Ecological Research Series EPA 660/3-75-009, April 1975; 61p

Water temperature: 22 °C ( ±1,0 )

Water hardness: 40 - 45 mg/l as CaCO<sub>3</sub>.

Alkalinity: 30 - 35 mg/l as CaCO<sub>3</sub>.

Dissolved oxygen: 9.0 mg/l (at time = zero)

pH levels: During test between 7.2 – 7.6

(2) American Public Health Association. 1980. Standard methods for the examination of water and wastewater. 15th ed. Washington DC 1134p.

Year: 1984

GLP: yes

Species:	Salmo gairdneri (42 ± 3.4 mm and 1.09 ± 0.28 g)
Report number:	Analytical Bio Chemistry lab # 32147
Type:	Static
Exposure period:	96 hour(s)
Test substance:	80% L(+) Lactic Acid
Nominal concentrations:	0, 32, 56, 100, 180 and 320 mg/l
Analytical monitoring:	yes
NOEC:	= 56 mg/l - measured/nominal
24h LC50:	= 150 mg/l - measured/nominal
48 h LC50:	= 130 mg/l - measured/nominal
96h LC50:	= 130 mg/l - measured/nominal

Method:

The static fish bioassay was conducted in five gallon glass vessels containing 15 litres reconstituted water. The study was conducted at the nominal concentrations of 32, 56, 100, 180 and 320 mg/l. Ten fish were exposed to each test concentration and control.

Water temperature:	12 °C ( ±1.0 )
Water hardness:	40 - 45 mg/l as CaCO <sub>3</sub> .
Alkalinity:	30 - 35 mg/l as CaCO <sub>3</sub> .
Dissolved oxygen:	During test between 6.1 – 9.2 mg/l
pH levels:	During test between 7.2 – 7.6

▪ **Acute Toxicity to Aquatic Invertebrates**

- Species: Daphnia magna (less than 24 hours old)
- Report number: TNO report R 91/294
- Type: Static
- Exposure period: 48 hour(s)
- Test substance: 80% L(+) Lactic Acid
- Nominal concentrations: 0, 32, 56, 100, 180, 320 and 560 mg/l
- Analytical monitoring: yes
- 48h NOEC: = 180 mg/l - calculated
- 48h EC50: = 240 mg/l - calculated
- 48h EC100: = 320 mg/l - calculated
- 24h EC 50: = 240mg /l - calculated
- Limit Test: no
- Method: OECD Guide-line 202
- Water temperature: 19.9 °C ( ±1 )
- Water hardness: 220 mg/l as CaCO<sub>3</sub>.
- Alkalinity: Unknown
- Total organic carbon: 2.0 mg/l
- Dissolved oxygen: > 7.9 mg/l
- pH levels: During test between 3.6 – 8.2
- Year: 1992
- GLP: yes
- Remark:  
test solutions are not neutralised. It is more than likely that the low pH values affected the mobility of the daphnia's.

- Species: Daphnia magna (less than 24 hours old)
- Report number: Analytical Bio Chemistry lab # 32148
- Type: Static
- Exposure period: 48 hour(s)
- Test substance: 80% L(+) Lactic Acid
- Nominal concentrations: 0, 100, 180, 320, 560 and 1000 mg/l
- Analytical monitoring: yes
- NOEC: = 320 mg/l - measured/nominal
- LC50 48h : = 750 mg/l - measured/nominal
- Limit Test: yes

Method:

Five concentrations in duplicate of the test compound with ten Daphnia per 250 ml glass beaker were used. The concentrations were a logarithmic series ranging from 100 to 1000 mg/l and included a control. The procedure for static bioassay as described in (1) and (2) below were used.

(1) Methods of acute toxicity with fish, Macro invertebrates and Amphibians. Stephan, CE, chairman. 1975.

Committee on Methods for toxicity tests with aquatic organisms. US EPA Ecol. Res. Ser. 660/3-75009.

(2) American Public Health Association. 1980. Standard methods for the examination of Water and wastewater. 15th ed. Washington DC. 1134p.

Water temperature: 20 °C ( ±2 )  
 Water hardness: 225 - 275 mg/l as CaCO<sub>3</sub>.  
 Alkalinity: 325 - 375 mg/l as CaCO<sub>3</sub>.  
 Dissolved oxygen: 7.1 – 8.5 mg/l (77 – 92 percent saturation at 20°C)  
 pH levels: During test between 4.3 – 8.5  
 Year 1984  
 GLP: yes



▪ **Toxicity to Aquatic Plants e.g. Algae**

- Report number: TNO 92/009  
 Species: Selenastrum capricornutum (Algae)  
 Endpoint: growth rate  
 Exposure period: 70 hour(s)  
 Test substance: 80% L(+) neutralised L(+) lactic acid  
 Analytical monitoring: yes  
 NOEC: = 1.9 mg/l - calculated  
 EC10: = 2.3 mg/l - calculated  
 EC50: = 3.5 mg/l - calculated  
 EC90 : = 5.4 mg/l - calculated  
 Method: OECD Guideline 201 "Algae, Growth Inhibition Test"  
 Water temperature: 23 °C ( ±1 )  
 Lighting conditions: 120 ± 20%  $\mu\text{mol.S}^{-1}.\text{m}^{-2}$ .  
 Composition of Growth medium:

NH <sub>4</sub> Cl	15 mg/l
MgCL <sub>2</sub> .6H <sub>2</sub> O	12 mg/l
CaCl <sub>2</sub> .2H <sub>2</sub> O	18 mg/l
MgSO <sub>4</sub> .7H <sub>2</sub> O	15 mg/l
KH <sub>2</sub> PO <sub>4</sub>	1.6 mg/l
FeCl <sub>3</sub> .6H <sub>2</sub> O	80 $\mu\text{g/l}$
Na <sub>2</sub> EDTA.2H <sub>2</sub> O	100 $\mu\text{g/l}$
H <sub>3</sub> BO <sub>3</sub>	185 $\mu\text{g/l}$
MnCl <sub>2</sub> .4H <sub>2</sub> O	415 $\mu\text{g/l}$
ZnCl <sub>3</sub>	3 $\mu\text{g/l}$
CoCl <sub>2</sub> .6H <sub>2</sub> O	1.5 $\mu\text{g/l}$
CuCl <sub>2</sub> .2H <sub>2</sub> O	0.01 $\mu\text{g/l}$
Na <sub>2</sub> MoO <sub>4</sub> .2H <sub>2</sub> O	7 $\mu\text{g/l}$
NaHCO <sub>3</sub>	150 $\mu\text{g/l}$

The pH of this medium after equilibration with air is approximately 8.

Water hardness: Not applicable  
 Dissolved oxygen: flasks were shaken (100 rpm); no data on dissolved oxygen.  
 pH levels: During test between 3.6 – 8.2  
 Year: 1992  
 GLP: yes

▪ **Toxicity to Micro-organisms e.g. Bacteria**

- Type: other: laboratory incubations  
Species: Escherichia coli (Bacteria)  
Exposure period: 20 minute(s)  
EC100 : = 15 - measured/nominal  
Test substance: combinations of 1.0-1.5% lactic acid with 0.1% sodium benzoate, or 0.1% hydrogen peroxide, or 0.005% glycerol monolaurate.

Result:

At 22C complete inactivation of E. coli O157:H7 was observed after 20 min. of exposure to 1.5% lactic acid plus 0.1% hydrogen peroxide.

Conclusion:

The mentioned treatment could potentially be used to inactivate or reduce E. coli O157:H7 populations on raw products

- Type: other: laboratory incubations on lean beef muscle discs
- Species: other bacteria: *Listeria monocytogenes*, *Yersinia enterocolitica*, *Salmonella typhimurium*, *E.coli*, *Campylobacter jejuni*, *Staphylococcus aureus*, *Pseudomonas fragi*, *Brochotrix thermosphacta*.
- Exposure period: 0 minute(s)
- Remark: Acid temperature (20 & 50 C) and concentration (1%, 3%) and initial numbers of contaminating bacteria (log CFU/cm<sup>2</sup> of 3-6) were the variables studied.
- Result: The bactericidal efficacy of lactic acid was often distinct for each organism. Bacterial numbers were maximally reduced with 3% acid at 55C. *S.aureus*: 1.4 log cycle; *P. fragi*: 2.3 log cycle; *B. thermosphacta*: 2.8 log cycle reduction.

## ***References Chapter 3: Ecotoxicity***

- Acute / Prolonged toxicity to fish
  - TNO report R 91/295; The acute toxicity of L(+)-lactic acid to Brachydanio Rerio (OECD 203).
  - Acute toxicity of L(+) lactic acid to Rainbow Trout (Salmo Gairdneri)  
Analytical Biochemistry Laboratories Inc.  
Columbia, MO  
1984
  - Acute toxicity of L(+) lactic acid to Bluegill Sunfish (Lepomis macrochirus)  
Analytical Biochemistry Laboratories Inc.  
Columbia, MO  
1984
- Acute toxicity to aquatic invertebrates
  - TNO-report R 91/ 294; The acute toxicity of L(+)-lactic acid to Daphnia Magna (OECD 202, 48h).
  - Acute toxicity of L(+) lactic acid to Daphnia Magna  
Analytical Biochemistry Laboratories Inc.  
Columbia, MO  
1984
- Toxicity to aquatic plants e.g. algae
  - TNO report R 92/009 ; Effect of L(+)-lactic acid on the growth of the alga Selenastrum Capricornutum (OECD 201).
- Toxicity to micro-organisms e.g. bacteria
  - Food Microbiology 16: 75-82 (1999), Venkitanarayanan K.S., Zhao T., Doyle M.P., "Inactivation of E.coli 0157:H7 by combinations of GRAS chemicals and temperature".
  - Greer G.G. and Dilts B.D., Factors affecting the susceptibility of meat borne pathogens and spoilage bacteria to organic acids.  
Food Research International 25: 355-362 (1992).

## **1. Toxicokinetics, Metabolism and Distribution**

- (L)-lactic acid is a natural functional metabolite in mammal, as mammalian fuel. According to the lactate shuttle concept, L-lactate represents a major means of distributing carbohydrate potential energy for oxidation and gluconeogenesis. The concept of a "lactate shuttle" (Brooks, 1998) is that during hard exercise, as well as other conditions of accelerated glycolysis, glycolic flux in muscle involves L-lactate formation regardless of the state of oxygenation. The production rate of endogenous (L)-lactate in the resting human is about 1.3 mol (70 kg/bw).24 h<sup>-1</sup> (= 117 g/day).

## 2. Acute oral toxicity

### LD50

Title	Acute oral LD50 study in rats																							
Species:	Charles River male/female rats																							
Report number	410-1369																							
Year	1984																							
GLP	Yes																							
Test substance & purity	L(+) lactic acid 80%, diluted with water.																							
Exposure / dosage	Oral dosage; 3,162 / 3,548 / 3,981 / 4,467 / 5,012 / 5,623 / 6,310 mg /kg bw																							
Exposure Time	One dosage																							
No. of Animals	55																							
Experimental design	Method:				EPA OPP 81-1																			
Observations	<p>The following mortalities were observed during the main study testing: dosage mg/kg: 3,162   3,548   3,981   4,467   5,012   5,623   6,310</p> <table><tr><td>males</td><td>--</td><td>--</td><td>--</td><td>1/5</td><td>3/5</td><td>4/5</td><td>5/5</td></tr><tr><td>females</td><td>1/5</td><td>2/5</td><td>5/5</td><td>5/5</td><td>5/5</td><td>5/5</td><td>5/5</td></tr></table> <p>All mortalities occurred after dosing on day 0 or in morning of day 1, except for one female dosed at 3,162 mg/kg that was found dead on the morning of day 2. The animals were sacrificed after 14-day observation. Consistent body weight gains were observed on days 7 and 14 for all surviving study animals.</p> <p>Lethargy, ataxia, prostration, irregular breathing, piloerection, squinting, lacrimation, salivation, crusty eyes and muzzle, loose stools, damp or yellow/brown stained fur, and moribund were abnormal clinical signs observed as early as 0 – 1 hour after dosing and as late as day 2. No other abnormal clinical signs were observed during the study.</p> <p>Abnormal necropsy findings were observed for all found dead animals and for the 4 surviving females dosed at 3,162 mg/kg.</p> <p>Abnormalities observed during necropsy of found dead animals included: discolored lungs; firm texture of lungs; green foci on one lung; erosion of stomachs; dark, black, brown and/or fluid contents of stomachs; black and/or brown discolored stomachs; a distended stomach with white mucosa; mucosal sloughing, ulceration and hemorrhage of the stomachs; discolored livers; white foci on livers; pale capsular areas, superficial erosion, or mottled livers; a discolored diaphragm; green-black or brown-black discolored kidneys; and red-brown exudate in the nasal and/or oral regions.</p> <p>Mottled lungs were observed during necropsy of 3 surviving animals dosed at 3,162 mg/kg and thickened stomachs were also observed during necropsy of 2 surviving animals of the same group. No other abnormalities were observed during necropsy of all study animals.</p>								males	--	--	--	1/5	3/5	4/5	5/5	females	1/5	2/5	5/5	5/5	5/5	5/5	5/5
males	--	--	--	1/5	3/5	4/5	5/5																	
females	1/5	2/5	5/5	5/5	5/5	5/5	5/5																	
Calculation of LD50	The oral LD50 value, the 95% confidence interval, the slope of the response curve, and correction factors for 0 and 100% observed responses were calculated for each sex using a method adapted from Litchfield and Wilcoxon. <sup>1</sup> Dose-response curves were prepared using the calculated LD50 data.																							
Conclusion	LD <sub>50</sub> between 3543 and 4936 mg/kg bw																							
Remark																								

<sup>1</sup> Lichtfield, J.T., Jr. and Wilcoxon, F., „A simplified method of evaluating Dose-effect experiments“, Journal of Pharmacology and Experimental Therapeutics, vol 96, 1949, pages 99-113.

**LD100**

Title	LD 100
Species:	10 rats
Concentration:	Oral dosage; dose was daily increased: 0.25 ml till 4.5 ml lactic acid 50%
Report number	
Year	
GLP	
Test substance & purity	L(+) lactic acid 80% water
Exposure	
Exposure Time	
No. of Animals	
Experimental design	
Observations	2 Rats died after dosing with 3 ml = 7500 mg/kg bw. The animals had a 15% reduction in bw in 1 week. A single administration of large doses did not result in changes in carbon dioxide content or pH of the blood, but a considerable decrease in the pH of the urine.
Conclusion	LD <sub>100</sub> : = 11250 mg/kg bw
Remark	

**Acute toxicity**

Title	Acute oral toxicity study in rats
Species:	Rats, Charles River
Report number	410-1353
Year	1983
GLP	Yes
Test substance & purity	L(+) lactic acid 80%, diluted with water
Exposure	5 mg / kg bodyweight
Exposure Time	One dosage
No. of Animals	10, 5 male and 5 female
Experimental design	EPA OPP 81-1. The duration of the study was 14 days. The animals were fasted overnight. The following morning, body weights were recorded, doses were calculated and a measured volume of test article was delivered to each animal by gavage in a single dose. Diet was returned to each animal approx 4 hours after test article administration. Observations for mortality and abnormal clinical signs were done twice daily. Body weights were recorded prior to test article administration on day 0, on day 7 and prior to sacrifice on day 14. Also final body weight were recorded prior to necropsy for animals found dead.
Observations	Body weight gains were observed for 3 surviving males and a small weight loss was observed for the fourth surviving animal. Body weight loss was observed for all the animals that were found dead. Lethargy and salivation were observed for all animals and crusty muzzle was observed for 9 animals on the day of dosing and as late as day 9 for one female. Other abnormal clinical signs observed for some animals on the day of dosing and as late as day 2 included ataxia, prostration, irregular breathing, noisy breathing, squinting, lacrimation, crusty eyes, crusty nose and body cool to touch. Other abnormal clinical signs observed prior to death of one female on day 10 included yellow/brown stained fur in the perianal region, abdomen swollen, no stools, and few stools. No other abnormal clinical signs were seen during the study.
Conclusion	Four males survived the 14-day duration of the study. One male and all females were found dead on the day of dosing (day 0), on day 1 or on day 10.
Remark	Test was done to establish clinical signs after single dose treatment

### 3. Acute Inhalation Toxicity

#### Acute Inhalation toxicity

Title	Acute Inhalation toxicity study
Species:	Rat Fischer 344 male/female
Concentration:	
Report number	I-7083.112
Year	1987
GLP	Yes
Test substance & purity	
Exposure	Aerosol containing 7,94 mg/L (nose only) a second group was exposed to air alone.
Exposure Time	4 hour(s)
No. of Animals	10
Experimental design	EPA OPP 81-3; The animals were observed for mortality and pharmacotoxic signs after exposure, at 1 and 3 hours following exposure and once daily after that for 14 days. Complete necropsis were performed on all animals on day 15 of the study. Histopathology was not performed
Observations	Animals were observed during exposure for signs of toxicity. Rapid breathing and eye tearing was observed in the treatment group. One and three hours after exposure, the treated and control groups had a hunched posture, red stained fur surrounding the eyes(tearing), ruffled fur, and appeared ungroomed with solid fur. Female rats exposed to the test substance appeared lethargic at one (2/5) and three hours (5/5). The two female rats that were lethargic after 1 hour also had a rapid, shallow breathing and appeared to be gasping after one and three hours. By 24 hours, most animals appeared normal and no unusual behaviour or appearance was observed for the remainder of the test period. However, of the treated female rats 4/5 had ruffled, ungroomed fur at 24 hours, and 3/5 had ruffled, ungroomed fur 2, 3 and 4 days after treatment. One female from the treatment group had hunched posture, rapid and shallow breathing, and slight tremors, but these signs were observed on day 5 post-treatment. One female from the treatment group died on day 8 post-treatment. No gross lesions were observed at necropsy.
Conclusion	Based on these results, the LC50 for L(+)Lactic Acid is greater than 7,94 mg/L.
Remark	



## 4. Acute Dermal Toxicity

- Type: Dermal LD50  
 Species: rabbit  
 Value: > 2000 mg/kg bw  
 Method: OECD Guide-line 402 "Acute dermal Toxicity"

### Acute dermal toxicity

Title	Acute dermal toxicity study in rabbits
Species:	Rabbits, New Zealand white, male and female
Report number	410-1354
Year	1983
GLP	Yes
Test substance & purity	L(+) lactic acid 80%, diluted with water
Exposure	2 g/kg for 2 hours of exposure
Exposure Time	2 g/kg for 2 hours of exposure
No. of Animals	10
Experimental design	EPA OPP 81-2; Approx 24 hours after clipping the skin were abraided sufficiently deep to penetrate the stratum corneum but not the dermis. After test substance application, the trunk of each animal was wrapped. After a 24 hour exposure period, each binder was removed and the test site of the animal was wiped to remove the remaing test article. Animals were observed for mortality and abnormal clinical signs hourly after dosing on day 0. Observations for mortality and abnormal clinical signs were done twice daily thereafter during the duration fo the study (14 days). On day 14 all animals were rendered unconscious and were exsanguinated prior to gross necropsy.
Observations	All animals survived the 14-days duration of the study and gained body weight. No abnormal clinical signs were observed during the study. Severe erythema and severe oedema were observed at the test sites of all animals after removal on day 1. Erythema decreased in severity for 3 animals on day 12 or 14, and was not observed for one female on day 14. Oedema decreased in severity for 8 animals and was not observed for one female on day 12 and for one male on day 14. Other dermal reactions observed at test sites included: blanching, necrosis, eschar formation, atonia, desquamation and denuded areas. Other dermal reactions observed at test site included: Blanching, necrosis, eschar formation, eschar peeled off, atonia, desquamation, fissures, denuded areas along abraided lines. Brown, crusted and raised discolorations of the treated skin were observed during necropsy of 3 males and 3 females. Multiple depressions in the treated skin were observed during necropsy of one of the same males, of 2 other males , and of one other female. A dark red focus was observed on the lung of one male. No other abnormalities were observed during necropsy of all males and 4 females, and no abnormalities were observed during necropsy of one female.
Conclusion	Dermal LD50 >2000 mg/kg in rabbits
Remark	

## 5. Skin Irritation

Title	Acute dermal irritation/corrosion test with lactic acid (88%) in albino rabbits
Species:	Rabbit
Concentration:	88 %
Report number	V 86.016
Year	1986
GLP	Yes
Test substance & purity	L(+) lactic acid 88%, diluted with water
Exposure	Occlusive
Exposure Time	4 hour(s)
No. of Animals	12
Experimental design	OECD Guide-line 404 "Acute Dermal Irritation/Corrosion"; One day prior to the experiment the hair was removed from the back and flanks of the animals. Six rabbits were treated on the intact skin and on the abraded skin. The abrasions were minor incisions of the stratum corneum, but not sufficient to disturb the underlying derma or to produce bleeding. An amount of 0,5 ml of the test material was brought onto the intact and the abraded skin under a surgical patch measuring 1 inch x 1 inch. After an exposure period of 4 hours the patches and the material applied were removed and the resulting skin reactions were evaluated by the method of Draize et.al. (J. Pharmacol. Exptl. Therap. 82 (1944) 377-390).
Observations	<p>After 4 hours the dermal effects generally observed in all rabbits consisted of very slight to slight ischemic necrosis, moderate to severe haemorrhages and slight or moderate oedema. After 28 hours the dermal effects observed generally consisted of very slight to slight ischemic necrosis, moderate haemorrhages, slight or moderate incrustation and slight oedema.</p> <p>During the course of the following two days ischemic necrosis, haemorrhages and oedema were no longer observed. The application sites generally became crater-shaped with a central sunken area which was moderately or severely encrusted, and a surrounding, raised border of non-necrotic skin showing well-defined erythema.</p> <p>After 7 days this picture had hardly changed, apart from clearance of erythema. The central sunken areas of the application sites generally showed moderate to severe incrustation.</p> <p>At the end of the observation period, after 3 weeks, some signs of healing were observed at the edges of the encrusted skin areas which had been in contact with the test material. In the new skin visible under the crust edges coming off from the treated skin, formation of scar tissue could be observed whereas hair growth was absent. There were no differences between reactions of the intact skin and those of abraded skin. This scar tissue formed already or to be formed is not considered a reversible skin alteration.</p>
Conclusion	On the basis of the results obtained, it can be concluded that lactic acid (88%) is severely irritating to the skin of albino rabbits and that lactic acid (88%) is corrosive to the skin of albino rabbits.
Remark	Other studies have shown that the skin of albino rabbit is not the appropriate animal model when addressing the effects of lactic acid on human skin. This result is therefore not used for the classification.

Title	Acute dermal irritation / corrosion study with PURAC BF S36 and PURAC BF S/30 in albino rabbits
Species:	New Zealand White albino rabbits
Concentration:	buffered lactic acid: BF S36 (38% l.a. + 38% sodium lactate, total 76%

	d.s.). BF S30 (60% l.a. + 20% sodium lactate, total 80% d.s.)
Report number	V 96.677
Year	1996
GLP	Yes
Test substance & purity	buffered lactic acid: BF S36 (38% l.a. + 38% sodium lactate, total 76% d.s.). BF S30 (60% l.a. + 20% sodium lactate, total 80% d.s.)
Exposure	Occlusive
Exposure Time	4 hours.
No. of Animals	3
Experimental design	<p>Four days prior to the start of the study, the hair was removed from the back and the flanks of the animals in a way to avoid abrasions. Each rabbit was treated simultaneously with the same compound.</p> <p>An amount of 0,5 ml of each test substance was distributed over a patch measuring 2.5 x 2.5 cm. The two patches were fixed to the application sites. Subsequently, the entire trunk of the rabbit was wrapped with self-adhesive tape.</p> <p>After a 4 hour exposure period, the test substance and patches were removed and the test sites were cleaned with moistened tissues.</p> <p>After a hour the resulting skin reactions were evaluated by the method of Draize et al (J. Pharmacol. Exp. Ther. 82 (1944) 377-390). Further skin readings were made at approximately 24, 48 and 72 hours after treatment.</p>
Observations	After 1-72 hours after treatment, no signs of skin irritation were observed in any of the three rabbits.
Conclusion	On the basis of the results obtained, it was concluded that, under the conditions of this study, PURAC BF36 and PURAC BF S30 are not irritating for the skin of rabbits after a 4 hour dermal contact period.
Remark	

Title	Primary dermal irritation study in rabbits
Species:	New Zealand White albino rabbits
Concentration:	80%
Report number	410-1355
Year	1983
GLP	Yes
Test substance & purity	L(+) lactic acid 80%, diluted with water
Exposure	Occlusive
Exposure Time	24 hours
No. of Animals	6, 3 male and 3 female
Experimental design	<p>Two test sites were prepared on each side of the spinal column in the thoracic region of each animal by closely clipping the hair.</p> <p>Approximately 24 hours after clipping, 0.5 ml of the test article was applied to each 2.5 cm square surgical gauze patch. 2 Application sites on each animal were abraded with a needle to penetrate the stratum corneum but not the dermis. Four patches containing the test article were then applied to each animal and held in place with gauze wrapping.</p> <p>After 24 hours exposure, the patches were removed. Each test site was gently wiped with gauze sponges moistened with water to remove the remaining test article.</p> <p>The skin conditions of each test site was evaluated for erythema, edema and other lesions at 30 and 60 minutes after test article removal. After 30-60 minutes evaluations all animals were euthanized due to severity of the dermal reactions observed.</p> <p>EPA OPP 81-5</p>
Observations	<ul style="list-style-type: none"> <li>- Severe erythema at 10 of the 12 abraded test sites (on 5 animals) and at 7 of the 12 intact test sites on 4 of the same animals. Moderate to severe erythema was seen at 2 remaining abraded and 5 remaining intact test sites.</li> <li>- Severe edema at 11 of 12 intact test sites and at 11 of 12 abraded test</li> </ul>

	<p>sites. Slight edema at one abraded test site and one intact test site on one animal.</p> <ul style="list-style-type: none"> <li>- Blanching at both abraded sites on each animal and at both intact sites at 5 animals.</li> <li>- Yellow-brown color of skin at all sites on 3 animals, at both abraded sites on one animal and at one intact site and both abraded sites on a fifth animal.</li> <li>- Red exudate at one intact site on one animal.</li> <li>- Skin missing at all sites of one animal, at one intact site and both abraded sites on one animal, and at one intact site or one abraded site on 2 other animals.</li> </ul> <p>No other dermal reactions were observed at the evaluations done at 30 to 60 minutes after test article removal. No abnormal clinical sign were observed and no mortalities occurred prior to sacrifice after 30 – 60 minute evaluations.</p>
Remark	Under the definition of CFR 49, 173.136, the product does not need to be classified

Title	Lactic acid Q88: A skin corrosivity test in guinea pigs
Species:	Guinea pigs, Dunkin Hartly strain
Concentration:	88%
Report number	235943
Year	1986
GLP	Yes
Test substance & purity	L(+) lactic acid 88%, diluted with water
Exposure	Occlusive
Exposure Time	Up to 4 hours
No. of Animals	6
Experimental design	<p>The animals were divided into 2 groups of 3. The hair was clipped from both flanks 24 hours before patch application. Care was taken to avoid abrading of the skin.</p> <p>In group 1, 0.5 ml of the test material was applied (at two sites) under 2 gauze patches each measuring 2.5 x 2.5 cm. The patches were applied to the intact skin and covered with Micropore tape. The whole trunk was then bound with Elastoplast elastic bandage to give a semi-occlusive covering. Group 2 animals were treated in the same way except that the test material was applied under one patch.</p> <p>For group 1 animals the first patch was removed 3 minutes after application and the skin was gently washed with water to remove any residual test material. Reactions were scored. One hour after application the second patch was removed, the skin washed and reactions scored.</p> <p>The patch for group 2 animals was left in place for 4 hours before removal and scoring.</p> <p>Skin reactions were scored for erythema and eschar formation and for oedema formation at patch removal, 1, 24, 48 and 72 hours after patch removal.</p>
Observations	<p>In group 1 ( 3 minutes and 1 hour exposure) no erythema, no eschar formation and no oedema formation was seen.</p> <p>In group 2 ( 4 hours exposure) very slight erythema (barely perceptible) and very slight oedema formation (barely perceptible) was seen at patch removal and after one hour.</p> <p>In group 2 no erythema, no eschar formation and no oedema formation was seen at 24, 48 and 72 hours after patch removal.</p>
Conclusion	It is concluded from the test results that lactic acid 88% is not corrosive to guinea pig skin. It is also noted that irritation after 4 hours exposure was transient and limited to very mild erythema.
Remark	

Title	Acute dermal irritation/corrosion study with lactic acid 88% in pigs
Species:	Healthy male, young pigs from Large white (GY) x Dutch Landrace (NL)
Concentration:	88%
Report number	V 87.405
Year	1987
GLP	Yes
Test substance & purity	L(+) lactic acid 88%, diluted with water
Exposure	Dermal, 0,5 ml test material per application site, occlusive.
Exposure Time	Up to 4 hours
No. of Animals	3
Experimental design	The hair was removed from the animals. The test material was brought into contact with three small separate areas of intact skin. Each of the application sites was covered with occlusive patch measuring 1 inch x 1 inch. The first patch was removed after 3 minutes, the second after 1 hour and the third after 4 hours. Immediately after removing the patches the treated skin areas were treated with lukewarm water, and one hour later the resulting skin reactions were evaluated by the method of Draize et.al. (J. Pharmacol. Exp. Ther. 82 (1944) 377-390). Further readings were made after 1 day and after 2, 3, 7, 14 and 21 days.
Observations	No dermal irritaton responses related to treatment with lactic acid (88%) were observed at the test sites of any animal during the 21-day observation period, etither after a 3 minute contact eperiod, after 60 minutes contact period or after a 4-hour contact period. After one day, superficial wounds were observed at the test site of pig no. 1 treated for 60 minutes and at the test site of pig no. 3 treated for 4 hours. In the same animals, similar injuries were also seen at the non-treated skin. After two days, the same test sites and non-treated skin areas in pigs 1 and 3 showed some slight small crusts. After three days these affects had cleared. These minor injuries were probably caused by shaving along the walls or the floor of the stable.
Conclusion	Under the conditions of this test it is concluded that lactic acid (88%) is not a primairy skin irritant to the pig skin.
<u>Remark</u>	TNO and other experts believe the pig to be a more appropriate and representative animal model than the albino rabbit, when addressing the effects of lactic acid on human skin.

## 6. Eye Irritation

Species	rabbit
Concentration:	20 % L(+) lactic acid
Year	1973
GLP	yes
Test substance & purity	20 % L(+) lactic acid in water
Experimental design	Journal Officiel de la Republique Francaise procedure; eyes were examined after 1 and 24 h and after 2, 3, 4, and 7 days with fluorescent staining.
Observations	In same study also 50% sodium lactate was tested, which is not irritating. Instilled at 20% and 10% provoked significant ocular irritation :Acute Ocular Irritation Index (AOII) was 39.50 resp. 31.17. Only for the 10% dilution these lesions were reversible, 7 days after instillation.
Conclusion	irritating
Remark	<b>From public literature, see references.</b>

Title	Chicken Enucleated eye test with three samples of lactic acid; an alternative to the Draize eye test.
Species:	Male or female chickens (ROSS, spring chickens) were used as eye donor.
Concentration:	
Report number	V96.157
Year	1996
GLP	Yes
Test substance & purity	<ul style="list-style-type: none"> <li>- Powder H60 (60% L(+) lactic acid and 40% calcium S lactate)</li> <li>- HS88, 88% L(+) lactic acid in water.</li> <li>- BF S36 L(+) lactic acid; L(+) lactic acid and sodium S lactae.</li> </ul>
Exposure	Corneal
Exposure Time	10 seconds
No. of Animals	10 eyes
Experimental design	<p>Within 2 hours after the kill the heads were carefully dissected and placed in a superfusion apparatus. The eyes are checked for damages. Corneal thickness was measured. Per test sample three eyes were selected for testing whereas one eye was rinsed with isotonic saline only and served as a control of the experimental conditions.</p> <p>At time t=0 0.03 ml / 0.03 g of the test sample was applied to the eye in such a way that the entire surface of the cornea was bathed with the test material. After a total exposure period of 10 seconds, the corneal surface was rinsed thoroughly with 20 ml of isotonic saline of ambient temperature. The control and the test eyes were examined for changes in corneal thickness (swelling), corneal opacity and fluorescein retention at 0, 30, 75, 120, 180 and 240 minutes after treatment.</p>
Observations	<ul style="list-style-type: none"> <li>- <u>powder H60</u> After treatment the thickness of the cornea of the eyes gradually increased considerably; a maximum mean corneal swelling of 17% was obtained at 240 minutes after treatment. In addition, moderate corneal opacity and moderate fluorescein retention by damaged epithelial cells were observed in the eyes. The categories defined for corneal swelling, corneal opacity and fluorescein retention were II, III and III</li> <li>- <u>PURAC lactic acid HS88</u> After treatment severe to complete corneal opacity was observed in the three test eyes, which hampered the measurement of corneal thickness</li> </ul>

	<p>at the 30, 75 and 120 minutes after treatment. At 180 and 240 minutes after treatment corneal thickness could be measured again and at 240 minutes a maximum corneal swelling of 28% was determined. All three eyes showed severe fluorescein retention by damaged epithelial cells. The categories defined for corneal swelling, corneal opacity and fluorescein retention were III, IV and IV</p> <ul style="list-style-type: none"> <li>- <u>PURAC lactic acid bufered BF S36</u> After treatment only a slight increase in cornela thickness was observed. A maximum corneal swelling of 6% was obtained 75 minutes after treatment. Very slight corneal opacity and slight fluorescein retention by damaged epithelial cells were observed in the test eyes. The categories defined for corneal swelling, corneal opacity and fluorescein retention were II, I and II</li> <li>- <u>Control eye</u> The control eye did not show any unusual effects.</li> </ul>
Conclusion	<ul style="list-style-type: none"> <li>- <u>powder H60</u>  Irritating to the eyes</li> <li>- <u>lactic acid HS88</u>  Severely irritating to the eyes</li> <li>- <u>lactic acid bufered BF S36</u>  Not irritating to the eye.</li> </ul>
Remark	The measurement of the cornela swelling in this test guarantees a highly objective parameter, which enables the investigator to discriminate the damaging effects of the test material very precisely, this in contrast to the conventional rabbit test which uses subjetive gross measurement only.

Species:	other
Concentration:	85 %
Experimental design	<u>Test condition:</u> in vitro using the Eytex Assay (Avon Products, Inc, 1995). Most of the formulations were tested undiluted. pH of formulations varies from 7.52 to 2.02.
Observations	Only formulation with pH 2.02 (face cream with 11.8% lactic acid 85%) was moderate severe irritant. The formulations with pH $\geq$ 5.3 were minimal irritant.
Conclusion	irritating
Remark	<b>From public literature, see references.</b>

## 7. Sensitisation

Title	Dermal sensitization study in guinea pigs (Buehler Test)
Species:	Female, Hartley guinea pigs
Concentration:	L(+) lactic acid 80%, diluted with water
Report number	480-2750
Year	1986
GLP	Yes
Test substance & purity	L(+) lactic acid 80%, diluted with water
Exposure	Dermal
Exposure Time	Two days for the range finding and 35 days for the main study.
No. of Animals	22 (2 for range finding, 10 for study test group and 10 for study naïve group)
Experimental design	The hair on the back or left flank (induction) and/or flank (range finding and challenge) of each animal was closely clipped. A 0.5 m sample of the test article and 0.5 ml samples of 30, 10 or 5 % suspensions of the test article in dionized water were placed on separate patches and applied to a range finding animal. The 100% test article was selected for induction.
Observations	<p>Preliminary range-finding trials revealed very slight erythema at the 100% concentration of the test article. No other dermal reactions were noted for the other concentrations (3, 10 and 30%). Therefor the 100% concentration of the test article was utilized in the main study testing for contact dermal sensitization potential.</p> <p>No mortalities occurred and all animals gained body weight. The test article (100%) produced very slight erythema at 3 sites and very slight edema at 1 site after the first induction. Erythema grades increased in severity after the second induction application. Due to the increase of severity of the reactions, the concentration of the test article was reduced to 30% and the induction site was changed to the left flank. Very slight erythema was noted after the fifth induction application. Grades ranging from very slight to severe erythema were noted from the seventh to the ninth induction applications.</p> <p>After the challenge application the test article (100%) produced grade 4 erythema in up to six test animals and in up to eight control animals. These reactions were considered irritation reactions, not sensitization reactions. The reactions seen in the control animals at the challenge were similar to the reactions seen for the test group animals and the test article was not considered to be a contact dermal sensitizer.</p>
Conclusion	The reactions seen (very slight to moderate erythema, very slight to moderate oedema) were considered to be irritant reactions, not sensitive reactions. The test article was not considered to be a dermal sensitizer.
Remark	

## 8. Repeated Dose Toxicity

Type	Sub-chronic
Species:	rat
Concentration:	4 ml lactic acid 10% on 20 g of meal
Exposure Time	Exposure period: 90 days
Experimental design	Route of administration: gavage Control Group: yes, concurrent no treatment Frequency of treatment: every day



Observations	No differences in appearance, gross observations at necropsy, or organ weights were observed between the test and control animals. Changes in blood carbon dioxide were slight.
Conclusion	
Remark	<b>From public literature, see references.</b>

Type	Sub-chronic
Species:	female Sprague-Dawley rat
Test substance & purity	formulation (face cream containing 0.25% of lactic acid 85%)
Exposure	dermal
Exposure Time	13 weeks treatment daily, 5 days/week Control Group: yes, concurrent no treatment
Experimental design	Doses: 886 mg/kg bw
Observations	No significant gross observations, with the exception of minimal skin irritation. Absolute brain weight and kidney-to-body weight ratios were increased for test animals. No lesions were observed at necropsy or at microscopic examination.
Conclusion	The formulation is safe in terms of cumulative toxicity. Based upon the exaggerated dose level used in this study for skin care products, dermal application is not likely to produce adverse effects under conditions of consumer use.LOAEL: 886 mg/kg
Remark	<b>From public literature, see references.</b>

Type	Sub-chronic
Species:	Fischer 344 male/female rat
Concentration:	experiment I: calcium lactate dissolved in drinking water (up to 5%). experiment II: up to 30% calcium lactate in diet.
Test substance & purity	Calcium lactate as a salt of lactic acid.
Exposure	Oral; 5, 2.5, 1.25, 0.6, 0.3 %
Exposure Time	Exposure period: 13 weeks
Experimental design	Control Group: yes, concurrent no treatment
Observations	a <10 % decrease in body weight gain. all animals survived. some haematological and biochemical parameters changed, but no severe lesions were found in microscopic examination in the experiment with calcium lactate mixed in the diet, the amount of calcium in the urine was significantly increased. Nephrocalcinosis and degeneration in kidneys observed. Indications that Nephrocalcinosis was dependent on the low Ca/Phosphorus ratio of the synthetic diet.
Remark	Lactic acid tested as its Calcium salt. From this study the lactate part is relevant, should be separated from effects of the soluble Calcium intake. <b>From public literature, see references.</b>

Type:	Sub-chronic
Species:	Syrian hamster male/female
Exposure	Group 1 (control):Diet 1, contains 20% sucrose as carcinogenic diet; pure water to drink. Group 2: diet 1, mixed with 0.057 ml lactic acid 80%; pure water to drink. Group 3: same diet 1, but water containing 0.050% v/v lactic acid
Exposure Time	14 weeks
Experimental design	daily ad libitum; animals of groups 2 & 3 ingest same amount lactic acid. Post exposure period: sacrificed and autopsy; also oral cavity (caries incidence) Control Group: yes, concurrent no treatment

Observations	three groups same growth and health. No significant differences were found in the incidence or extent of carious lesions among the three groups.
Conclusion	dietary lactic acid did not play any important role in development or progress of dental caries.
<u>Remark</u>	pH of diet 1 is 5.55, of diet 2 is 5.12. pH of pure water is 6.8 and of water + lactic acid is 3.1 . <b>From public literature, see references.</b>

## 9. Genetic Toxicity 'in Vitro'

Type:	Ames test
Species:	
Concentration:	0.5, 1.0, and 2.0 microliter lactic acid/plate
Report number	
Year	
GLP	
Test substance & purity	
Exposure	
Exposure Time	
No. of Animals	
Experimental design	Salmonella/microsome test (Ames test) and chromosomal aberration test in vitro reverse mutation assays, and Chinese hamster fibroblast cell line S. typhimurium strains TA97, TA98, TA100, TA104 Metabolic activation: with and without
Observations	negative
Conclusion	
Remark	<b>From public literature, see references.</b>

Type:	Chinese hamster ovary K1 cells, chromosomal aberration tests, and the pH relationship of the medium and clastogenic activity was examined.
Concentration:	8-35 mM
Experimental design	Cells were maintained in Ham's F12 medium, supplemented with 10% foetal calf serum. Cytotoxic Concentration: 14-35 mM, when pH was $\leq 5.8$ Metabolic activation: with and without
Observations	When the culture medium was first acidified by the lactic acid dose and then neutralised to pH 6.4 or when medium is containing 30 mM HEPES as buffer, lactic acid was non-clastogenic. Pseudo-positive reactions are seen as a result of non-physiological low pH.
Conclusion	lactic acid was non-clastogenic
Remark	<b>From public literature, see references.</b>

Type	review on several mutagenicity studies with lactic acid and some lactates.
Concentration:	various
Experimental design	various Metabolic activation: with and without
Conclusion	negative
Remark	the result of 11 studies is reviewed <b>From public literature, see references.</b>

## 10. Genetic Toxicity ' in Vivo '

Due to the natural nature of L(+) lactic acid and the relative low contribution of "outside L(+) lactic acid" to the human metabolism, in vivo genotoxicity studies will not be required.

### ▪ **Carcinogenicity**

Title	
Species:	Fischer 344 male/female rat
Concentration:	2.5 or 5 % Calcium lactate in the drinking water. Mean total Calcium lactate intake for males was 329.4 g, resp. 625.4 g; for females 237.7 g, resp. 412.1 g.
GLP	yes
Test substance & purity	The Calcium salt of lactic acid was tested
Exposure	drinking water daily, ad lib
Exposure Time	2 years
Experimental design	Control Group: yes, concurrent no treatment
Observations	Autopsy on rats that died during study and those killed at the end. Examination macro-and microscopically for presence of non-neoplastic and neoplastic lesions
Conclusion	Negative
Remark	Lactic acid tested as its Calcium salt. From this study the lactate intake is relevant, should be separated from the Calcium effects of a soluble Calcium salt. <b>From public literature, see references.</b>

Species:	female rabbit
Concentration:	0.1-0.2 g/kg bw (5 months), and 0.1-0.7 g/kg bw (13 months)
Exposure	drinking water twice daily
Exposure Time	5 or 13 months
Observations	No tumors were reported after 5 or 16 months. Further details not provided.
Conclusion	negative
Remark	<b>From public literature, see references.</b>

## 11. Toxicity to Fertility

The nature of the compound (part of human metabolism) does make toxicity studies to fertility not necessary

## 12. Developmental Toxicity/Teratogenicity

Species:	female CD-1 mouse
Exposure Time	gestational days 6-15 Duration of test: 10 days
Experimental design	Administration: gavage Frequency of treatment: daily Doses: 570 mg/kg bw/day Control Group:yes, concurrent no treatment
Observations	Lactic acid was neither maternotoxic nor embryofetotoxic when given orally to mice at 570 mg/kg bw/day on gestation days 6-15.
Conclusion	NOAEL Maternal Toxicity: $\geq 570$ mg/kg bw NOAEL Teratogenicity: $\geq 570$ ml/kg bw
Remark	<b>From public literature, see references.</b>

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**HPV Data Set**

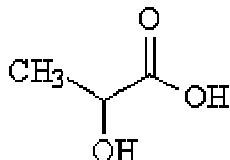
**LACTIC ACID**

**CAS # 50-21-5**

**Dossier number 50215**

## Substance information

<b>CAS No.</b>	50-21-57
<b>EINECS Name:</b>	lactic acid
<b>EC No.</b>	200-018-0
<b>TSCA Name:</b>	Propanoic acid, 2-hydroxy-, (2S)-
<b>Molecular Formula</b>	C <sub>3</sub> H <sub>6</sub> O <sub>3</sub>



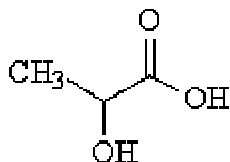
<b>IUPAC Name:</b>	LACTIC ACID
<b>Mol. Weight:</b>	90
<b>Physical status:</b>	liquid or white crystals
<b>Colour:</b>	colourless to slightly yellow
<b>Odour:</b>	nearly odourless

Most of the lactic acid nowadays is supplied in the L(+) form, the natural form.

PURAC only manufactures the L(+) form.

Substance information for the L(+) lactic acid is as follows:

<b>CAS No.</b>	79-33-4
<b>EINECS Name:</b>	L(+) lactic acid
<b>EC No.</b>	201-196-2
<b>TSCA Name:</b>	Propanoic acid, 2-hydroxy-, (2S)-
<b>Molecular Formula</b>	C <sub>3</sub> H <sub>6</sub> O <sub>3</sub>



<b>IUPAC Name:</b>	L(+) LACTIC ACID
<b>Mol. Weight:</b>	90
<b>Physical status:</b>	liquid or white crystals
<b>Colour:</b>	colourless to slightly yellow
<b>Odour:</b>	nearly odourless

This HPV data file typically contains information about L(+) lactic acid.

# Chapter 1                      Physico-chemical Data

## ▪ **Melting Point**

Value:                                      ≤ 54 degree C  
Decomposition:                      no at < 110 degree C  
Sublimation:                              no  
Method:                                      other  
GLP:    no data  
Test substance:                      100% crystalline pure product used

## ▪ **Melting Point of the anhydrous form of the racemic mixture**

Kraft and Dyes found for the racemic lactic acid the melting point 18°C., Borsook, Huffman and Liu reported 16.8°C and more recently Lockwood, Yoder and Zienty reported values of up to 28-33°C. When one of the optically active isomers is present in excess, the melting point is higher; this indicates that racemic lactic acid forms a eutectic mixture.

Ref: ***Lactic Acid: properties and chemistry of lactic acid and derivatives by C.H. Holten, 1971, page 21***

## ▪ **Boiling Point / Vapour pressure**

Boiling point:                              ca. 258 degree C at 1000 hPa  
Vapour pressure:                      ca. .0041 hPa at 20 degree C  
Decomposition:                              yes  
Method:                                      other  
GLP:    no data  
Remark:                                      solutions can polymerise on boiling

It is not possible to determine this value very accurate, because lactic acid tends to polymerise by a polycondensation reaction.

The boiling point of lactic acid at difference vapour pressures according to

- Aspen Database
- PURAC In-house Pure compound database

**Table** Boiling points of lactic acid at different pressures (Aspen Database)

Pressure (mbar)	Boiling temperature (°C)
1	79.6
5	101.9

10	112.8
50	141.8
100	156.2
200	172.1
400	189.8
600	201.0
800	209.4
1013	216.6

The vapour pressures of lactic acid as a function of the boiling temperature according to the Aspen formula (PLXANT).

$$PLXANT = e^{C1 + \frac{C2}{C3+T} + C4 \cdot T + C5 \cdot \ln T + C6 \cdot T^2}$$

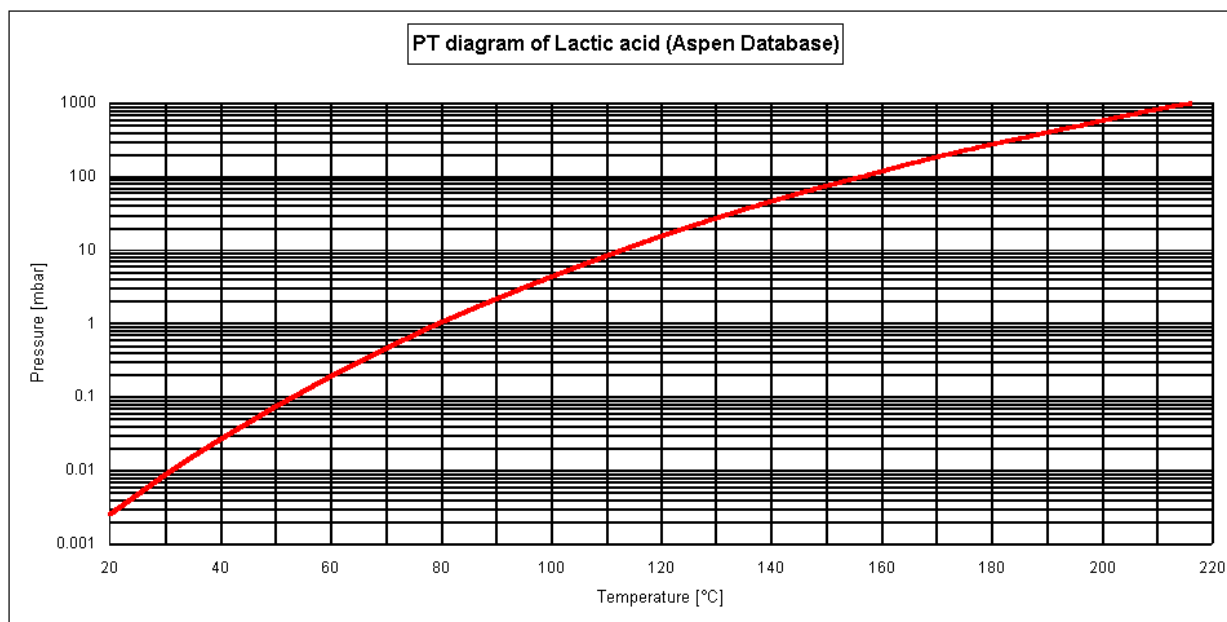
: Vapour pressure (N/m<sup>2</sup> or Pa)

PLXANT

C1, C2 C3, C4, C5, C6, C7 : Regression coefficients for chemical compound

T : Boiling temperature (K)

**Figure** Vapour Pressure diagram



The regression coefficients are:

C1 = 225.19

C5 = -28.816

C2 = -18757

C6 = 0.000012998

C3 = 0

C7 = 2

C4 = 0

G.P.v.Lieshout measured the vapour pressure of lactic acid between 0 and 180 °C.

**Table** Boiling points of lactic acid at different pressures

Temperature (°C)	Vapour Pressure (mbar)
0	0.0005
10	0.0015
20	0.0041
30	0.0107
40	0.0258
50	0.0588
60	0.1273
70	0.2629
80	0.5198
90	0.9877
100	1.8093
110	3.2042
120	5.5004
130	9.1736
140	14.8959
150	23.5938
160	36.5163
170	55.3114
180	82.1111

The vapour pressures of lactic acid as a function of the boiling temperature were fitted with the Aspen formula (PLXANT). The estimated atmospheric boiling point is about 258 °C.

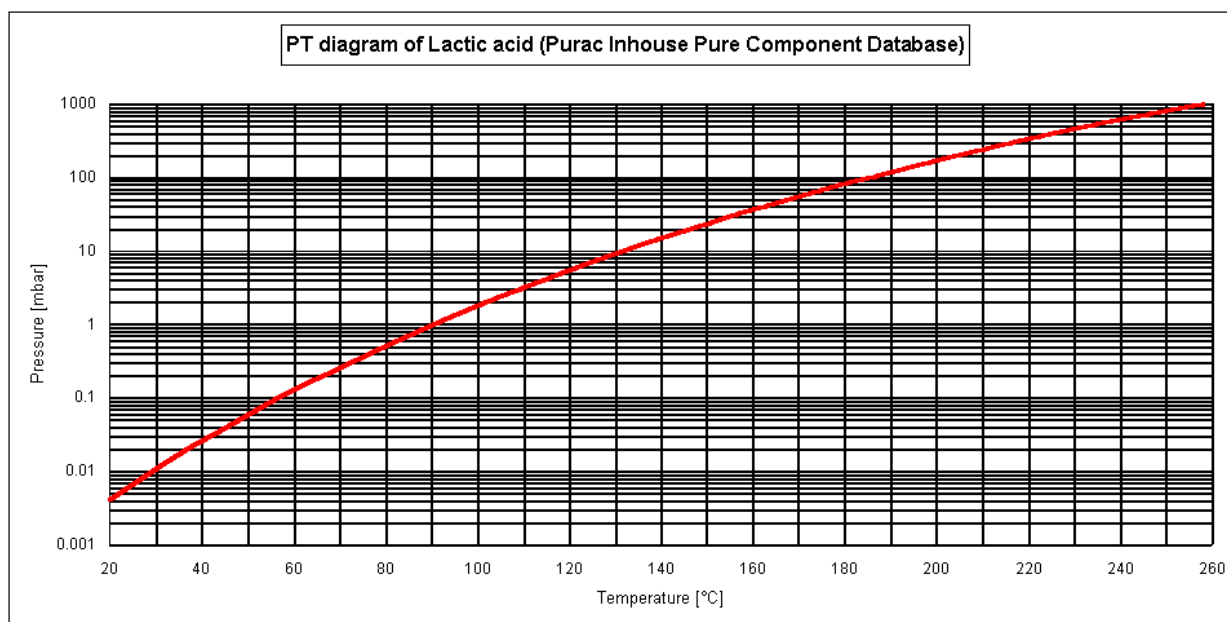
$$\text{PLXANT} = e^{\frac{C1 + \frac{C2}{C3+T}}{C4 \cdot T + C5 \cdot \ln T + C6 \cdot T^2}}$$

: Vapour pressure (N/m<sup>2</sup> or Pa)

PLXANT

C1, C2 C3, C4, C5, C6, C7

: Regression coefficients for chemical compound



: Boiling temperature (K)

Figure: Vapour Pressure diagram

The regression coefficients are:

C1 = 32.4649                      C5 = -0.376037

C2 = -8835.5                    C6 = 0

C3 = 0                            C7 = 0

C4 = -0.003654

## ▪ Density

Type: density  
 Value:  $\geq 1.2255 \text{ g/cm}^3$  at 20 degree C  
 Method: other  
 GLP: yes  
 Test substance: Lactic acid 80%

Holten describes the liquid density of lactic acid solutions in water.

**Table** Liquid Densities of aqueous lactic acid solutions at different temperatures.

Concentration (wt%)	Temperature (°C)										
	0	10	20	30	40	50	60	70	80	90	100
0	1.000 4	0.999 7	0.997 9	0.995 3	0.991 9	0.987 8	0.983 1	0.977 8	0.972 0	0.965 8	0.959 1
10	1.026 9	1.025 1	1.022 3	1.018 8	1.014 5	1.009 7	1.004 2	0.998 3	0.991 9	0.985 1	0.977 9
20	1.053 5	1.050 5	1.046 8	1.042 4	1.037 3	1.031 7	1.025 6	1.019 0	1.012 0	1.004 7	0.997 0
30	1.080 2	1.076 2	1.071 5	1.066 2	1.060 4	1.054 0	1.047 2	1.040 0	1.032 5	1.024 6	1.016 4
40	1.106 9	1.101 9	1.096 4	1.090 2	1.083 6	1.076 6	1.069 2	1.061 4	1.053 3	1.044 9	1.036 3
50	1.133 3	1.127 5	1.121 1	1.114 2	1.106 9	1.099 2	1.091 2	1.082 9	1.074 3	1.065 5	1.056 4
60	1.159 1	1.152 4	1.145 3	1.137 7	1.129 9	1.121 6	1.113 1	1.104 3	1.095 3	1.086 1	1.076 6
70	1.183 5	1.176 2	1.168 5	1.160 4	1.152 1	1.143 4	1.134 5	1.125 4	1.116 0	1.106 5	1.096 7
80	1.206 1	1.198 3	1.190 2	1.181 8	1.173 1	1.164 2	1.155 0	1.145 6	1.136 0	1.126 3	1.116 3
90	1.226 3	1.218 3	1.210 0	1.201 4	1.192 5	1.183 4	1.174 1	1.164 6	1.155 0	1.145 1	1.135 1
95	1.235 4	1.227 4	1.219 0	1.210 4	1.201 5	1.192 4	1.183 1	1.173 6	1.163 9	1.154 1	1.144 1
100	1.243 8	1.235 7	1.227 4	1.218 8	1.209 9	1.200 8	1.191 6	1.182 1	1.172 5	1.162 7	1.152 7

## Liquid Density of pure lactic acid

The liquid density of pure lactic acid at difference temperatures according to:

- Aspen Database
- PURAC In-house Pure Component Database



**Table** Densities of lactic acid at different temperatures (ASPEN Database)

Temperature (°C)	Density (g/ml)
20	1.2255
30	1.2163
40	1.2069
50	1.1974
60	1.1878
70	1.1780
80	1.1681
90	1.1580
100	1.1478
110	1.1374
120	1.1268
130	1.1160
140	1.1051
150	1.0939
160	1.0825

The liquid densities of lactic acid as a function of the temperature according to the Aspen equation for liquid density (DNLDIP):

$$\text{DNLDIP} = \frac{C1}{C2^{1+(1-\frac{T}{C3})^{C4}}}$$

DNLDIP : Liquid density (kmole/m<sup>3</sup>)

C1, C2 C4 : Regression coefficients for chemical compound

C3 : Critical temperature (K)

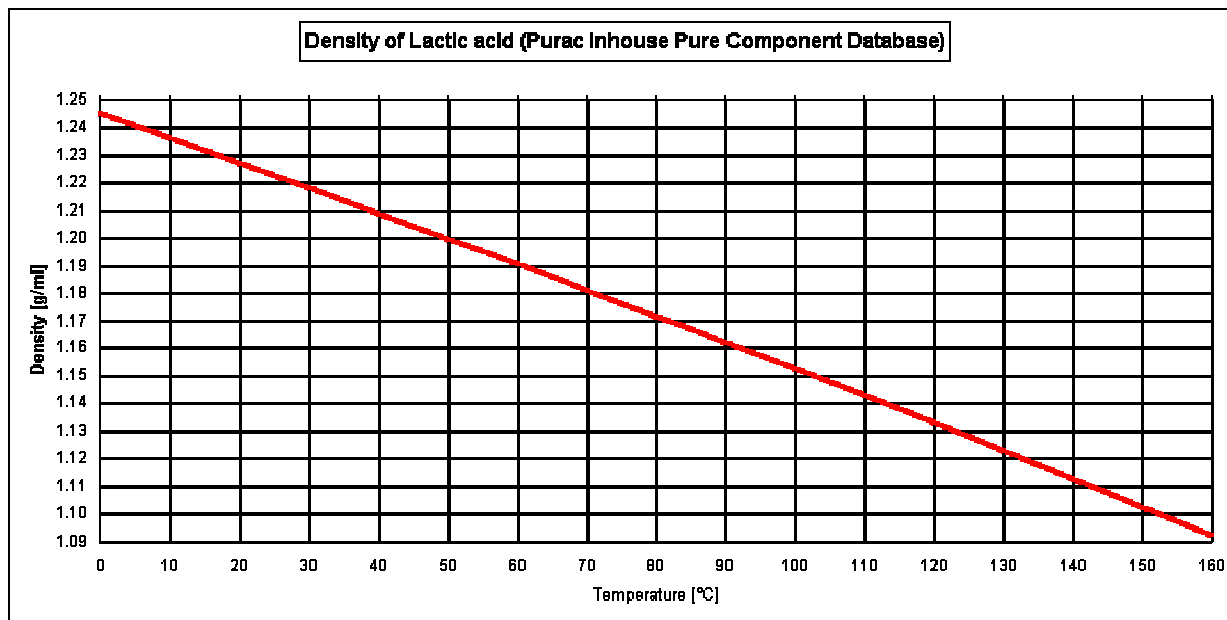
T : Temperature (K)

The regression coefficients are:

C1 = 0.9798                      C3 = 675

C2 = 0.24593                    C4 = 0.2333

### Liquid density diagram



**Table** Densities of lactic acid at different temperatures (Purac In-house Pure Component Database)

Temperature (°C)	Density (g/ml)
0	1.2453
10	1.2364
20	1.2274
30	1.2183
40	1.2092
50	1.2000
60	1.1907
70	1.1813
80	1.1719
90	1.1623
100	1.1527
110	1.1429
120	1.1330
130	1.1231
140	1.1130
150	1.1028
160	1.0924

The liquid densities of lactic acid as a function of the temperature according to the Aspen equation for liquid density (DNLDIP):

$$\text{DNLDIP} = \frac{C1}{C2^{1 + (1 - \frac{T}{C3})^{C4}}}$$

: Liquid density (kmole/m<sup>3</sup>)

DNLDIP

C1, C2 C4 : Regression coefficients for chemical compound

C3 : Critical temperature (K)

T : Temperature (K)

The regression coefficients are:

C1 = 1.61248 C3 = 729.9

C2 = 0.314479 C4 = 0.328317

#### ▪ **Partition Coefficient**

Partition Coeff.: octanol-water

log Pow: ca. -.62 at 20 degree C

Method: OECD Guide-line 117 "Partition Coefficient (n-octanol/water),  
HPLC Method"

Year: 1987

GLP: yes

#### ▪ **Solubility in water**

Value: ca. 100 vol% at 25 degree C

pH value: ca. 1.2

pKa: 3.68 at 25 degree C

Descr.: of very high solubility

Method: other

GLP: no data

Deg. product: not measured

Stable: yes

Remark: completely soluble at 25 degrees C

J.v.Krieken determined the phase diagram of lactic acid/ water. Due to practical problems the diagram wasn't completed, but the missing part was extrapolated.

**Table** Solubility of monomeric lactic acid in water

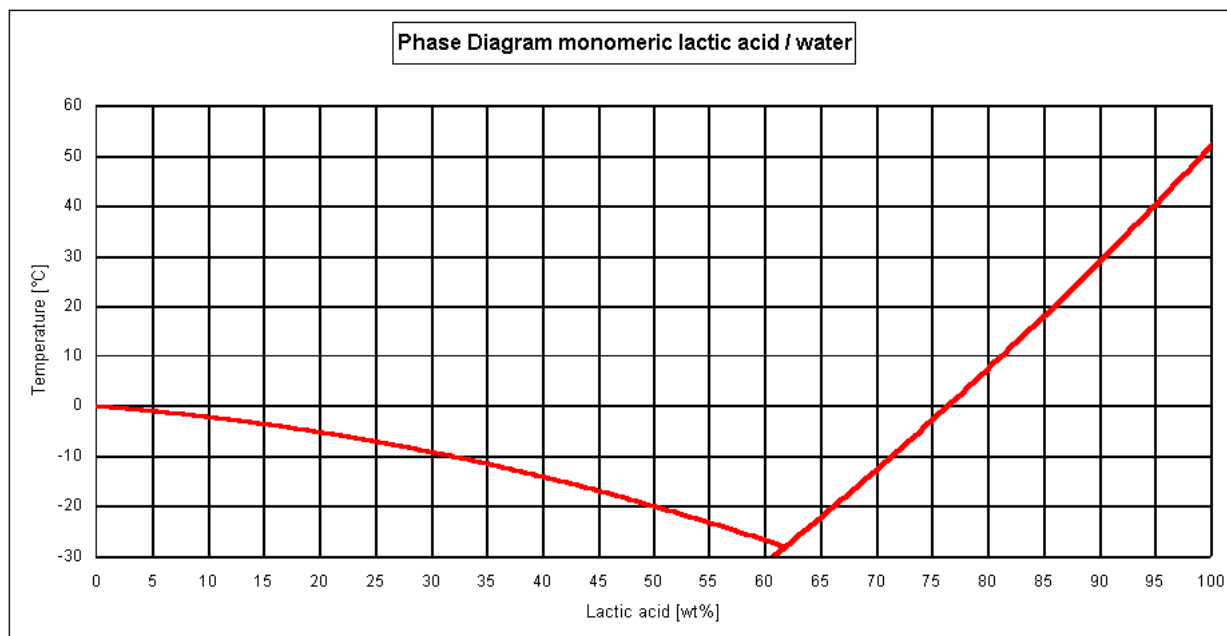
Temperature (°C)	Lactic acid (wt%)
-20	65.9
-15	68.8
-10	71.5
-5	74.2
0	76.7
5	78.6
10	81.3
15	83.5
20	86.1
25	87.6
35	92.7
40	95.1

**Table** Freezing point data of monomeric lactic acid in water

Temperature (°C)	Lactic acid (wt%)
------------------	-------------------

0	0
-5	18.6
-10	32.5
-15	41.9
-21	51.7

The eutectic point of lactic acid/water was calculated by extrapolation and is ca. 61.7 wt% lactic acid and -27.1 °C.



**Figure** Phase Diagram

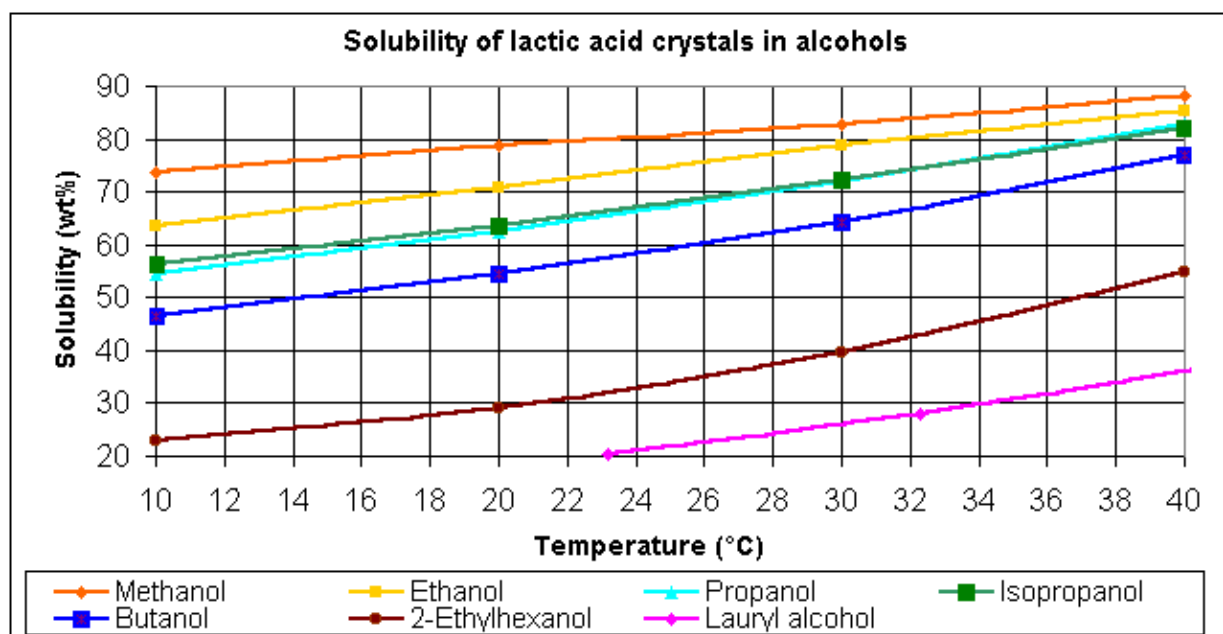
### Solubility of lactic acid in other solvents

J.v. Krieken investigated the solubility of pure lactic acid in a range of solvents by stirring the specific solvent with an excess of (S)-lactic acid crystals at a specific temperature. After separation of the solids from the liquid, the clear liquid was analysed on free acidity.

**Table** solubility of lactic acid in a range of solvents

Solvent	Solubility (wt%) of solution			
	10 °C	20 °C	30 °C	40 °C
Methanol	73.8	78.6	82.8	88.1
Ethanol	63.6	70.9	78.7	85.2
1-Propanol	54.5	62.4	71.7	82.7
2-Propanol	56.1	63.4	72.2	82.2
1-Butanol	46.3	54.5	64.3	77.1
2-Ethyl-1-hexanol	22.9	29.0	39.6	54.9
Cyclohexane				0.04
Hexane	< 0.01	< 0.01	< 0.01	0.02

Toluene	0.06	0.11	0.24	0.50
Ethyl lactate	37.0	45.9	57.2	72.8
Butyl lactate	27.7	35.7	47.3	64.8
2-Ethylhexyl lactate	15.5	20.9	30.9	46.7
Ethyl acetate	27.1	39.9	56.2	72.9
Diethyl ether	23.8	38.7	59.8	
Diisopropyl ether	6.4	9.2	15.9	44.4
Tetrahydrofuran	58.4	65.1	72.4	82.9
Dichloromethane	0.59	1.02	2.7	
Chloroform	0.31	0.67	1.67	59.7
2-Butanone	40.7	52.9	67.3	81.6
Acetone	53.4	61.4	71.5	82.9



## ▪ Viscosity

Test type: other: ASPEN database ; PURAC internal.  
Value: ca. 53.52 mPa s (dynamic) at 20 degree C.

The liquid viscosity of lactic acid at difference temperatures according to.

- Aspen Database.
- PURAC In-house Pure Component Database.

The liquid viscosity of aqueous lactic acid solutions in water.

- 0 - 88 wt% of Lactic Acid.
- 90 - 110 wt% Lactic Acid.

**Table** Viscosity of lactic acid at different temperatures (Aspen Database).

Temperature (°C)	Viscosity (cP)
20	53.52
30	33.26
40	21.29
50	14.01
60	9.44
70	6.51
80	4.59
90	3.29
100	2.40
110	1.78
120	1.34
130	1.03
140	0.79
150	0.62
160	0.49

The liquid viscosity of lactic acid as a function of the temperature according to the Aspen equation for liquid viscosity (MULDIP):

$$\text{MULDIP} = e^{C1 + \frac{C2}{T} + C3 \ln T + C4 \cdot T^{C5}}$$

: Liquid viscosity (N\*s/m<sup>2</sup> or Pa.s)

MULDIP

C1, C2 C3, C4, C5 : Regression coefficients for chemical compound

T : Temperature (K)

The regression coefficients are:

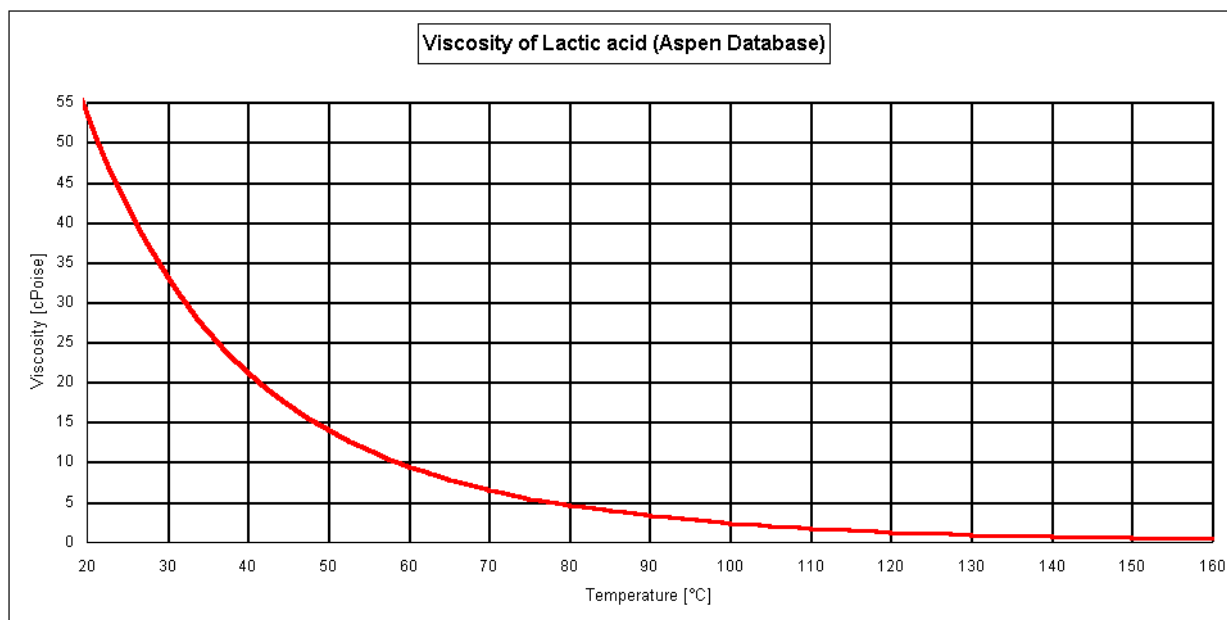
C1 = -14.403

C2 = 4097.9

C3 = -0.4407

C4 = 0

C5 = 0



**Figure:** Liquid Viscosity diagram

Table 12 Viscosity of lactic acid at different temperatures (Purac In-house Pure Component Database) <sup>2</sup>

Temperature (°C)	Viscosity (cP)
20	1342.33
30	579.14
40	280.59
50	150.73
60	88.80
70	56.81
80	39.14
90	28.83
100	22.54

The liquid viscosity of lactic acid as a function of the temperature according to the Aspen equation for liquid viscosity (MULDIP):

MULDIP:                    Liquid viscosity ( $\text{N}\cdot\text{s}/\text{m}^2$  or  $\text{Pa}\cdot\text{s}$ )

C1, C2, C3, C4, C5: Regression coefficients for chemical compound

T:                            Temperature (K)

The regression coefficients are:

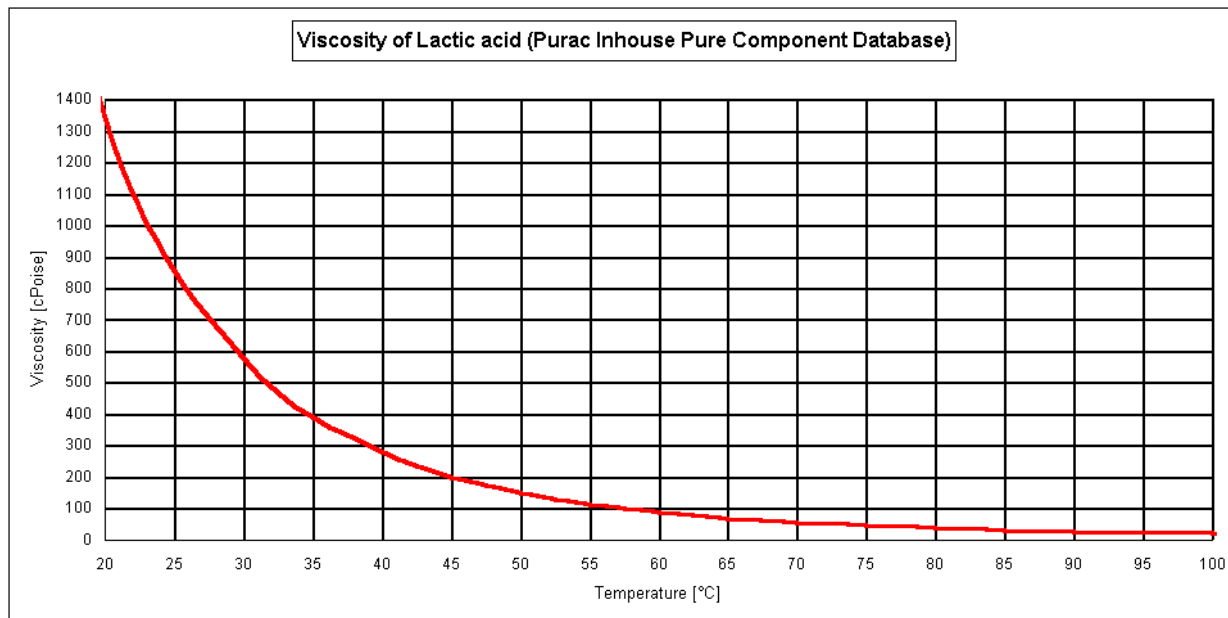
C1:                        421.094

C2:                        25091.4

C3:                        59.1119

C4:                        0

C5:                        0



**Figure:** Liquid Viscosity diagram

## ***References Chapter 1: Physico chemical data***

- National Chemical Inventories  
Codex NCINF5  
ISSN: 1089-6279
- Handbook of Chemistry and Physics  
WEAST  
66th Edition
- ASPEN database
- PURAC internal databank



## **Chapter 2 Environmental Fate and Pathways**

### ▪ **Photodegradation**

The photochemical oxidisation of lactic acid is discussed in "Lactic acid properties and chemistry of lactic acid and derivatives by C.H. Holten (1971)". The first observation that lactic acid is photosensitive was made in 1910 by Berthelot and Gaudechon, who irradiated calcium lactate and ethyl lactate with ultraviolet rays. They observed decomposition with the formation of gas containing carbon monoxide, carbon dioxide, hydrogen and methane.

Recognized method, i.e. OECD: Modelling conducted, no guideline studies used.

Method: Estimated Programs Interface (EPIWIN V3.05, Atmospheric Oxidation Program v 1.90). Model executed in October 2002.

Results / observations: The AOP component of EPIWIN was used to calculate the rate of photodegradation for L(+) lactic acid.

SMILES : O=C(O)C(O)C

CHEM : Propanoic acid, 2-hydroxy-, (S)-

MOL FOR: C3 H6 O3

MOL WT : 90.08

----- SUMMARY (AOP v1.90): HYDROXYL RADICALS -----

Hydrogen Abstraction = 5.2598 E-12 cm<sup>3</sup>/molecule-sec

Reaction with N, S and -OH = 0.6600 E-12 cm<sup>3</sup>/molecule-sec

Addition to Triple Bonds = 0.0000 E-12 cm<sup>3</sup>/molecule-sec

Addition to Olefinic Bonds = 0.0000 E-12 cm<sup>3</sup>/molecule-sec

Addition to Aromatic Rings = 0.0000 E-12 cm<sup>3</sup>/molecule-sec

Addition to Fused Rings = 0.0000 E-12 cm<sup>3</sup>/molecule-sec

OVERALL OH Rate Constant = 5.9198 E-12 cm<sup>3</sup>/molecule-sec

HALF-LIFE = 1.807 Days (12-hr day; 1.5E6 OH/cm<sup>3</sup>)

HALF-LIFE = 21.682 Hrs

----- SUMMARY (AOP v1.90): OZONE REACTION -----

\*\*\*\*\* NO OZONE REACTION ESTIMATION \*\*\*\*\*

(ONLY Olefins and Acetylenes are Estimated)

Experimental Database: NO Structure Matches

### ▪ **Stability in Water**

Lactic Acid (88% and 60% aqueous solutions) were investigated.

"The kinetics of degradation of Lactic Acid was done at elevated temperature, since the decomposition rates of lactic acid, (..) were too slow to obtain kinetic data within

reasonable time. At the condition studied (25, 40, 80 and 120 degrees centigrade) the decompositions of these compounds followed apparent first order kinetics because the mean correlation coefficient was above 0.980.

Lactic acid was very stable in aqueous solutions at 80 degrees centigrade (less than 30% decomposition after 175 at 80 degrees) and degradation was not different when combined with the various excipients tested. The shelf lives determined for lactic acid ranged from 79 years when combined with isopropyl palmitate to 98 years when combined with sorbic acid.

#### ▪ **Transport between Environmental Compartments**

Type: other: see free texts RM.

Remark: Lactic acid is not volatile and it has a high biodegradation rate. Therefor transport between compartments is no issue for this compound.

No experimental data are available on fugacity. However, calculation using the sotware program EPI Suite version 3.1 gave the following information which supports the statement given above:

Level III Fugacity Model:			
	Mass Amount (%)	Half-Life (hr)	Emissions (kg/hr)
Air	3.16	43.4	1000
Water	46.3	208	1000
Soil	50.5	208	1000
Sediment	0.0691	832	0
Persistence Time: 222 hr			

#### • **Biodegradation**

Value 50% degradation after 5 days and 67% after 20 days.

Breakdown Product: It is to be expected that L(+) lactic acid will be taken up into the metabolism of the bacteria in the activated sludge. Degradation products will therefor be CO<sub>2</sub> and H<sub>2</sub>O.

Method: BOD (Biochemical Oxygen Demand) and COD (Chemical Oxygen Demand) determinations were carried out for L(+) lactic acid usig the method described in teh Dutch guidenlines "water determination of biochemical oxgen demand after n days (BODn)" (NEN 6634) and "Water determination of chemical oxygen demand (COD)" (NEN 6633) respectively, these methods are similar to those referred in the EC Test Guidelines C.8 and C.9 Two concentrations (2 mg/L and 4 mg/L) were tested. An oculum was prepared from activated sludge. Its microbial activity appeared to be sufficient although the control substance glucose and glutamic acid had a BOD5 of slightly less than the required value of 4.00 ± 0.75 mg O<sub>2</sub>/L

## ***References Chapter 2: Environmental fate and pathways***

- Lactic Acid: properties and chemistry of lactic acid and derivatives by C.H. Holten, 1971, page 38
- TNO report R 92/018: BOD and COd of L(+) lactic acid according to EC test guidelines C.8 and C.9
- The ecotoxicity and the biodegradability of lactic acid, alkyl esters and lactate salts  
C.T. Bowmer et.al.  
Chemosphere, volume 37, No 7, pp 1317 - 1333, 1998
- Stability of lactic acid and glycolic acid in aqueous systems subjected to acid hydrolysis and thermal decomposition.  
M.M. de Villiers et.al.  
Journal of the society of cosmetic chemists, 48, 165-174 (August 1998)
- Handbook of Chemistry and Physics  
WEAST  
66th Edition

## **Chapter 3      Ecotoxicity**

### ▪ ***Acute/Prolonged Toxicity to Fish***

- Species: Brachydanio rerio (2.1 ±0.17 cm and 0.07 ±0.018 g)
- Report number: TNO R 91/295
- Type: Semistatic
- Exposure period: 48 hour(s)
- Test substance: 80% L(+) Lactic Acid
- Nominal concentrations: 0, 100, 180, 320, 560 and 1000 mg/l
- Analytical monitoring: yes
- 48h LC50: = 320 - calculated
- Limit Test: no
- Method: OECD Guideline 203 "Fish, Acute Toxicity Test"
- Water temperature: Between 24.0 and 24.7 °C
- Water hardness: About 220 mg/l expressed as CaCO<sub>3</sub>.
- Alkalinity: Unknown
- Total Organic content: 2.0 mg/l
- Dissolved oxygen: > 7.3 mg/l
- pH levels: During test between 3.2 and 8.2
- Year: 1992
- GLP: yes
- Remark: test solutions are not neutralised. It is more than likely that the low pH value affected the survival of the fishes.

- Species: Brachydanio rerio (2.1 ±0.17 cm and 0.07 ±0.018 g)  
Report number: TNO R 91/295  
Type: Semistatic  
Exposure period: 96 hour(s)  
Test substance: 80% L(+) Lactic Acid  
Nominal concentrations: 0, 100, 180, 320, 560 and 1000 mg/l  
Analytical monitoring: yes  
96h NOEC: = 320 mg/l - measured/nominal  
96h LC50: = 320 mg/l - calculated  
96h LC100: = 560 mg/l - calculated  
Method: OECD Guideline 203 "Fish, Acute Toxicity Test"  
Water temperature: Between 24.0 and 24.7 °C  
Water hardness: About 220 mg/l expressed as CaCO<sub>3</sub>.  
Alkalinity: Unknown  
Total Organic content: 2.0 mg/l  
Dissolved oxygen: > 7.3 mg/l  
pH levels: During test between 3.2 and 8.1  
Year: 1992  
GLP: yes  
Remark: test solutions are not neutralised. It is more than likely that the low pH value affected the survival of the fishes.

- Species: Lepomis macrochirus (24± 2.3 mm and 0.37 ± 0.15 g)  
Report number: Analytical Bio Chemistry lab # 32146  
Type: Static  
Exposure period: 96 hour(s)  
Test substance: 80% L(+) Lactic Acid  
Nominal concentrations: 0, 56, 100, 180, 320 and 560 mg/l  
Analytical monitoring: yes  
NOEC: = 56 mg/l - measured/nominal  
24 h LC50: = 140 mg/l - measured/nominal  
48 h LC50: = 130 mg/l - measured/nominal  
96h LC50: = 130 mg/l - measured/nominal  
Limit Test: yes

Method: :

Ten fish were exposed to each test concentration and control.

The procedures for static bioassay described in (1) and (2) below were used in this experiment.

(1) Committee on methods for Toxicity Tests with Aquatic Organisms (C.E. Stephan chairman). 1975.

Methods for acute toxicity tests with fish, macro invertebrates and amphibians.

Environmental Protection Agency, Ecological Research Series EPA 660/3-75-009, April 1975; 61p

Water temperature: 22 °C ( ±1,0 )

Water hardness: 40 - 45 mg/l as CaCO<sub>3</sub>.

Alkalinity: 30 - 35 mg/l as CaCO<sub>3</sub>.

Dissolved oxygen: 9.0 mg/l (at time = zero)

pH levels: During test between 7.2 – 7.6

(2) American Public Health Association. 1980. Standard methods for the examination of water and wastewater. 15th ed. Washington DC 1134p.

Year: 1984

GLP: yes

Species:	Salmo gairdneri (42 ± 3.4 mm and 1.09 ± 0.28 g)
Report number:	Analytical Bio Chemistry lab # 32147
Type:	Static
Exposure period:	96 hour(s)
Test substance:	80% L(+) Lactic Acid
Nominal concentrations:	0, 32, 56, 100, 180 and 320 mg/l
Analytical monitoring:	yes
NOEC:	= 56 mg/l - measured/nominal
24h LC50:	= 150 mg/l - measured/nominal
48 h LC50:	= 130 mg/l - measured/nominal
96h LC50:	= 130 mg/l - measured/nominal

Method:

The static fish bioassay was conducted in five gallon glass vessels containing 15 litres reconstituted water. The study was conducted at the nominal concentrations of 32, 56, 100, 180 and 320 mg/l. Ten fish were exposed to each test concentration and control.

Water temperature:	12 °C ( ±1.0 )
Water hardness:	40 - 45 mg/l as CaCO <sub>3</sub> .
Alkalinity:	30 - 35 mg/l as CaCO <sub>3</sub> .
Dissolved oxygen:	During test between 6.1 – 9.2 mg/l
pH levels:	During test between 7.2 – 7.6

▪ **Acute Toxicity to Aquatic Invertebrates**

- Species: Daphnia magna (less than 24 hours old)
- Report number: TNO report R 91/294
- Type: Static
- Exposure period: 48 hour(s)
- Test substance: 80% L(+) Lactic Acid
- Nominal concentrations: 0, 32, 56, 100, 180, 320 and 560 mg/l
- Analytical monitoring: yes
- 48h NOEC: = 180 mg/l - calculated
- 48h EC50: = 240 mg/l - calculated
- 48h EC100: = 320 mg/l - calculated
- 24h EC 50: = 240mg /l - calculated
- Limit Test: no
- Method: OECD Guide-line 202
- Water temperature: 19.9 °C ( ±1 )
- Water hardness: 220 mg/l as CaCO<sub>3</sub>.
- Alkalinity: Unknown
- Total organic carbon: 2.0 mg/l
- Dissolved oxygen: > 7.9 mg/l
- pH levels: During test between 3.6 – 8.2
- Year: 1992
- GLP: yes
- Remark:  
test solutions are not neutralised. It is more than likely that the low pH values affected the mobility of the daphnia's.



- Species: Daphnia magna (less than 24 hours old)
- Report number: Analytical Bio Chemistry lab # 32148
- Type: Static
- Exposure period: 48 hour(s)
- Test substance: 80% L(+) Lactic Acid
- Nominal concentrations: 0, 100, 180, 320, 560 and 1000 mg/l
- Analytical monitoring: yes
- NOEC: = 320 mg/l - measured/nominal
- LC50 48h : = 750 mg/l - measured/nominal
- Limit Test: yes

Method:

Five concentrations in duplicate of the test compound with ten Daphnia per 250 ml glass beaker were used. The concentrations were a logarithmic series ranging from 100 to 1000 mg/l and included a control. The procedure for static bioassay as described in (1) and (2) below were used.

(1) Methods of acute toxicity with fish, Macro invertebrates and Amphibians. Stephan, CE, chairman. 1975.

Committee on Methods for toxicity tests with aquatic organisms. US EPA Ecol. Res. Ser. 660/3-75009.

(2) American Public Health Association. 1980. Standard methods for the examination of Water and wastewater. 15th ed. Washington DC. 1134p.

Water temperature: 20 °C ( ±2 )

Water hardness: 225 - 275 mg/l as CaCO<sub>3</sub>.

Alkalinity: 325 - 375 mg/l as CaCO<sub>3</sub>.

Dissolved oxygen: 7.1 – 8.5 mg/l (77 – 92 percent saturation at 20°C)

pH levels: During test between 4.3 – 8.5

Year 1984

GLP: yes

▪ **Toxicity to Aquatic Plants e.g. Algae**

- Report number: TNO 92/009  
 Species: Selenastrum capricornutum (Algae)  
 Endpoint: growth rate  
 Exposure period: 70 hour(s)  
 Test substance: 80% L(+) neutralised L(+) lactic acid  
 Analytical monitoring: yes  
 NOEC: = 1.9 mg/l - calculated  
 EC10: = 2.3 mg/l - calculated  
 EC50: = 3.5 mg/l - calculated  
 EC90 : = 5.4 mg/l - calculated  
 Method: OECD Guideline 201 "Algae, Growth Inhibition Test"  
 Water temperature: 23 °C ( ±1 )  
 Lighting conditions: 120 ± 20%  $\mu\text{mol.S}^{-1}.\text{m}^{-2}$ .  
 Composition of Growth medium:

NH <sub>4</sub> Cl	15 mg/l
MgCL <sub>2</sub> .6H <sub>2</sub> O	12 mg/l
CaCl <sub>2</sub> .2H <sub>2</sub> O	18 mg/l
MgSO <sub>4</sub> .7H <sub>2</sub> O	15 mg/l
KH <sub>2</sub> PO <sub>4</sub>	1.6 mg/l
FeCl <sub>3</sub> .6H <sub>2</sub> O	80 $\mu\text{g/l}$
Na <sub>2</sub> EDTA.2H <sub>2</sub> O	100 $\mu\text{g/l}$
H <sub>3</sub> BO <sub>3</sub>	185 $\mu\text{g/l}$
MnCl <sub>2</sub> .4H <sub>2</sub> O	415 $\mu\text{g/l}$
ZnCl <sub>3</sub>	3 $\mu\text{g/l}$
CoCl <sub>2</sub> .6H <sub>2</sub> O	1.5 $\mu\text{g/l}$
CuCl <sub>2</sub> .2H <sub>2</sub> O	0.01 $\mu\text{g/l}$
Na <sub>2</sub> MoO <sub>4</sub> .2H <sub>2</sub> O	7 $\mu\text{g/l}$
NaHCO <sub>3</sub>	150 $\mu\text{g/l}$

The pH of this medium after equilibration with air is approximately 8.

Water hardness: Not applicable  
 Dissolved oxygen: flasks were shaken (100 rpm); no data on dissolved oxygen.  
 pH levels: During test between 3.6 – 8.2  
 Year: 1992  
 GLP: yes

▪ **Toxicity to Micro-organisms e.g. Bacteria**

- Type: other: laboratory incubations  
Species: Escherichia coli (Bacteria)  
Exposure period: 20 minute(s)  
EC100 : = 15 - measured/nominal  
Test substance: combinations of 1.0-1.5% lactic acid with 0.1% sodium benzoate, or 0.1% hydrogen peroxide, or 0.005% glycerol monolaurate.

Result:

At 22C complete inactivation of E. coli O157:H7 was observed after 20 min. of exposure to 1.5% lactic acid plus 0.1% hydrogen peroxide.

Conclusion:

The mentioned treatment could potentially be used to inactivate or reduce E. coli O157:H7 populations on raw products

- Type: other: laboratory incubations on lean beef muscle discs
- Species: other bacteria: *Listeria monocytogenes*, *Yersinia enterocolitica*, *Salmonella typhimurium*, *E.coli*, *Campylobacter jejuni*, *Staphylococcus aureus*, *Pseudomonas fragi*, *Brochotrix thermosphacta*.
- Exposure period: 0 minute(s)
- Remark: Acid temperature (20 & 50 C) and concentration (1%, 3%) and initial numbers of contaminating bacteria (log CFU/cm<sup>2</sup> of 3-6) were the variables studied.
- Result: The bactericidal efficacy of lactic acid was often distinct for each organism. Bacterial numbers were maximally reduced with 3% acid at 55C. *S.aureus*: 1.4 log cycle; *P. fragi*: 2.3 log cycle; *B. thermosphacta*: 2.8 log cycle reduction.

## ***References Chapter 3: Ecotoxicity***

- Acute / Prolonged toxicity to fish
  - TNO report R 91/295; The acute toxicity of L(+)-lactic acid to Brachydanio Rerio (OECD 203).
  - Acute toxicity of L(+) lactic acid to Rainbow Trout (Salmo Gairdneri)  
Analytical Biochemistry Laboratories Inc.  
Columbia, MO  
1984
  - Acute toxicity of L(+) lactic acid to Bluegill Sunfish (Lepomis macrochirus)  
Analytical Biochemistry Laboratories Inc.  
Columbia, MO  
1984
- Acute toxicity to aquatic invertebrates
  - TNO-report R 91/ 294; The acute toxicity of L(+)-lactic acid to Daphnia Magna (OECD 202, 48h).
  - Acute toxicity of L(+) lactic acid to Daphnia Magna  
Analytical Biochemistry Laboratories Inc.  
Columbia, MO  
1984
- Toxicity to aquatic plants e.g. algae
  - TNO report R 92/009 ; Effect of L(+)-lactic acid on the growth of the alga Selenastrum Capricornutum (OECD 201).
- Toxicity to micro-organisms e.g. bacteria
  - Food Microbiology 16: 75-82 (1999), Venkitanarayanan K.S., Zhao T., Doyle M.P., "Inactivation of E.coli 0157:H7 by combinations of GRAS chemicals and temperature".
  - Greer G.G. and Dilts B.D., Factors affecting the susceptibility of meat borne pathogens and spoilage bacteria to organic acids.  
Food Research International 25: 355-362 (1992).

## **1. Toxicokinetics, Metabolism and Distribution**

- (L)-lactic acid is a natural functional metabolite in mammal, as mammalian fuel. According to the lactate shuttle concept, L-lactate represents a major means of distributing carbohydrate potential energy for oxidation and gluconeogenesis. The concept of a "lactate shuttle" (Brooks, 1998) is that during hard exercise, as well as other conditions of accelerated glycolysis, glycolic flux in muscle involves L-lactate formation regardless of the state of oxygenation. The production rate of endogenous (L)-lactate in the resting human is about 1.3 mol (70 kg/bw).24 h<sup>-1</sup> (= 117 g/day).

## 2. Acute oral toxicity

### LD50

Title	Acute oral LD50 study in rats																							
Species:	Charles River male/female rats																							
Report number	410-1369																							
Year	1984																							
GLP	Yes																							
Test substance & purity	L(+) lactic acid 80%, diluted with water.																							
Exposure / dosage	Oral dosage; 3,162 / 3,548 / 3,981 / 4,467 / 5,012 / 5,623 / 6,310 mg /kg bw																							
Exposure Time	One dosage																							
No. of Animals	55																							
Experimental design	Method:				EPA OPP 81-1																			
Observations	<p>The following mortalities were observed during the main study testing: dosage mg/kg: 3,162   3,548   3,981   4,467   5,012   5,623   6,310</p> <table><tr><td>males</td><td>--</td><td>--</td><td>--</td><td>1/5</td><td>3/5</td><td>4/5</td><td>5/5</td></tr><tr><td>females</td><td>1/5</td><td>2/5</td><td>5/5</td><td>5/5</td><td>5/5</td><td>5/5</td><td>5/5</td></tr></table> <p>All mortalities occurred after dosing on day 0 or in morning of day 1, except for one female dosed at 3,162 mg/kg that was found dead on the morning of day 2. The animals were sacrificed after 14-day observation. Consistent body weight gains were observed on days 7 and 14 for all surviving study animals.</p> <p>Lethargy, ataxia, prostration, irregular breathing, piloerection, squinting, lacrimation, salivation, crusty eyes and muzzle, loose stools, damp or yellow/brown stained fur, and moribund were abnormal clinical signs observed as early as 0 – 1 hour after dosing and as late as day 2. No other abnormal clinical signs were observed during the study.</p> <p>Abnormal necropsy findings were observed for all found dead animals and for the 4 surviving females dosed at 3,162 mg/kg.</p> <p>Abnormalities observed during necropsy of found dead animals included: discolored lungs; firm texture of lungs; green foci on one lung; erosion of stomachs; dark, black, brown and/or fluid contents of stomachs; black and/or brown discolored stomachs; a distended stomach with white mucosa; mucosal sloughing, ulceration and hemorrhage of the stomachs; discolored livers; white foci on livers; pale capsular areas, superficial erosion, or mottled livers; a discolored diaphragm; green-black or brown-black discolored kidneys; and red-brown exudate in the nasal and/or oral regions.</p> <p>Mottled lungs were observed during necropsy of 3 surviving animals dosed at 3,162 mg/kg and thickened stomachs were also observed during necropsy of 2 surviving animals of the same group. No other abnormalities were observed during necropsy of all study animals.</p>								males	--	--	--	1/5	3/5	4/5	5/5	females	1/5	2/5	5/5	5/5	5/5	5/5	5/5
males	--	--	--	1/5	3/5	4/5	5/5																	
females	1/5	2/5	5/5	5/5	5/5	5/5	5/5																	
Calculation of LD50	The oral LD50 value, the 95% confidence interval, the slope of the response curve, and correction factors for 0 and 100% observed responses were calculated for each sex using a method adapted from Litchfield and Wilcoxon. <sup>1</sup> Dose-response curves were prepared using the calculated LD50 data.																							
Conclusion	LD <sub>50</sub> between 3543 and 4936 mg/kg bw																							
Remark																								

<sup>1</sup> Lichtfield, J.T., Jr. and Wilcoxon, F., „A simplified method of evaluating Dose-effect experiments“, Journal of Pharmacology and Experimental Therapeutics, vol 96, 1949, pages 99-113.

**LD100**

Title	LD 100
Species:	10 rats
Concentration:	Oral dosage; dose was daily increased: 0.25 ml till 4.5 ml lactic acid 50%
Report number	
Year	
GLP	
Test substance & purity	L(+) lactic acid 80% water
Exposure	
Exposure Time	
No. of Animals	
Experimental design	
Observations	2 Rats died after dosing with 3 ml = 7500 mg/kg bw. The animals had a 15% reduction in bw in 1 week. A single administration of large doses did not result in changes in carbon dioxide content or pH of the blood, but a considerable decrease in the pH of the urine.
Conclusion	LD <sub>100</sub> : = 11250 mg/kg bw
Remark	

**Acute toxicity**

Title	Acute oral toxicity study in rats
Species:	Rats, Charles River
Report number	410-1353
Year	1983
GLP	Yes
Test substance & purity	L(+) lactic acid 80%, diluted with water
Exposure	5 mg / kg bodyweight
Exposure Time	One dosage
No. of Animals	10, 5 male and 5 female
Experimental design	EPA OPP 81-1. The duration of the study was 14 days. The animals were fasted overnight. The following morning, body weights were recorded, doses were calculated and a measured volume of test article was delivered to each animal by gavage in a single dose. Diet was returned to each animal approx 4 hours after test article administration. Observations for mortality and abnormal clinical signs were done twice daily. Body weights were recorded prior to test article administration on day 0, on day 7 and prior to sacrifice on day 14. Also final body weight were recorded prior to necropsy for animals found dead.
Observations	Body weight gains were observed for 3 surviving males and a small weight loss was observed for the fourth surviving animal. Body weight loss was observed for all the animals that were found dead. Lethargy and salivation were observed for all animals and crusty muzzle was observed for 9 animals on the day of dosing and as late as day 9 for one female. Other abnormal clinical signs observed for some animals on the day of dosing and as late as day 2 included ataxia, prostration, irregular breathing, noisy breathing, squinting, lacrimation, crusty eyes, crusty nose and body cool to touch. Other abnormal clinical signs observed prior to death of one female on day 10 included yellow/brown stained fur in the perianal region, abdomen swollen, no stools, and few stools. No other abnormal clinical signs were seen during the study.
Conclusion	Four males survived the 14-day duration of the study. One male and all females were found dead on the day of dosing (day 0), on day 1 or on day 10.
Remark	Test was done to establish clinical signs after single dose treatment



### 3. Acute Inhalation Toxicity

#### Acute Inhalation toxicity

Title	Acute Inhalation toxicity study
Species:	Rat Fischer 344 male/female
Concentration:	
Report number	I-7083.112
Year	1987
GLP	Yes
Test substance & purity	
Exposure	Aerosol containing 7,94 mg/L (nose only) a second group was exposed to air alone.
Exposure Time	4 hour(s)
No. of Animals	10
Experimental design	EPA OPP 81-3; The animals were observed for mortality and pharmacotoxic signs after exposure, at 1 and 3 hours following exposure and once daily after that for 14 days. Complete necropsis were performed on all animals on day 15 of the study. Histopathology was not performed
Observations	Animals were observed during exposure for signs of toxicity. Rapid breathing and eye tearing was observed in the treatment group. One and three hours after exposure, the treated and control groups had a hunched posture, red stained fur surrounding the eyes(tearing), ruffled fur, and appeared ungroomed with solid fur. Female rats exposed to the test substance appeared lethargic at one (2/5) and three hours (5/5). The two female rats that were lethargic after 1 hour also had a rapid, shallow breathing and appeared to be gasping after one and three hours. By 24 hours, most animals appeared normal and no unusual behaviour or appearance was observed for the remainder of the test period. However, of the treated female rats 4/5 had ruffled, ungroomed fur at 24 hours, and 3/5 had ruffled, ungroomed fur 2, 3 and 4 days after treatment. One female from the treatment group had hunched posture, rapid and shallow breathing, and slight tremors, but these signs were observed on day 5 post-treatment. One female from the treatment group died on day 8 post-treatment. No gross lesions were observed at necropsy.
Conclusion	Based on these results, the LC50 for L(+)Lactic Acid is greater than 7,94 mg/L.
Remark	

## 4. Acute Dermal Toxicity

- Type: Dermal LD50  
 Species: rabbit  
 Value: > 2000 mg/kg bw  
 Method: OECD Guide-line 402 "Acute dermal Toxicity"

### Acute dermal toxicity

Title	Acute dermal toxicity study in rabbits
Species:	Rabbits, New Zealand white, male and female
Report number	410-1354
Year	1983
GLP	Yes
Test substance & purity	L(+) lactic acid 80%, diluted with water
Exposure	2 g/kg for 2 hours of exposure
Exposure Time	2 g/kg for 2 hours of exposure
No. of Animals	10
Experimental design	EPA OPP 81-2; Approx 24 hours after clipping the skin were abraided sufficiently deep to penetrate the stratum corneum but not the dermis. After test substance application, the trunk of each animal was wrapped. After a 24 hour exposure period, each binder was removed and the test site of the animal was wiped to remove the remaing test article. Animals were observed for mortality and abnormal clinical signs hourly after dosing on day 0. Observations for mortality and abnormal clinical signs were done twice daily thereafter during the duration fo the study (14 days). On day 14 all animals were rendered unconscious and were exsanguinated prior to gross necropsy.
Observations	All animals survived the 14-days duration of the study and gained body weight. No abnormal clinical signs were observed during the study. Severe erythema and severe oedema were observed at the test sites of all animals after removal on day 1. Erythema decreased in severity for 3 animals on day 12 or 14, and was not observed for one female on day 14. Oedema decreased in severity for 8 animals and was not observed for one female on day 12 and for one male on day 14. Other dermal reactions observed at test sites included: blanching, necrosis, eschar formation, atonia, desquamation and denuded areas. Other dermal reactions observed at test site included: Blanching, necrosis, eschar formation, eschar peeled off, atonia, desquamation, fissures, denuded areas along abraided lines. Brown, crusted and raised discolorations of the treated skin were observed during necropsy of 3 males and 3 females. Multiple depressions in the treated skin were observed during necropsy of one of the same males, of 2 other males , and of one other female. A dark red focus was observed on the lung of one male. No other abnormalities were observed during necropsy of all males and 4 females, and no abnormalities were observed during necropsy of one female.
Conclusion	Dermal LD50 >2000 mg/kg in rabbits
Remark	

## 5. Skin Irritation

Title	Acute dermal irritation/corrosion test with lactic acid (88%) in albino rabbits
Species:	Rabbit
Concentration:	88 %
Report number	V 86.016
Year	1986
GLP	Yes
Test substance & purity	L(+) lactic acid 88%, diluted with water
Exposure	Occlusive
Exposure Time	4 hour(s)
No. of Animals	12
Experimental design	OECD Guide-line 404 "Acute Dermal Irritation/Corrosion"; One day prior to the experiment the hair was removed from the back and flanks of the animals. Six rabbits were treated on the intact skin and on the abraded skin. The abrasions were minor incisions of the stratum corneum, but not sufficient to disturb the underlying derma or to produce bleeding. An amount of 0,5 ml of the test material was brought onto the intact and the abraded skin under a surgical patch measuring 1 inch x 1 inch. After an exposure period of 4 hours the patches and the material applied were removed and the resulting skin reactions were evaluated by the method of Draize et.al. (J. Pharmacol. Exptl. Therap. 82 (1944) 377-390).
Observations	<p>After 4 hours the dermal effects generally observed in all rabbits consisted of very slight to slight ischemic necrosis, moderate to severe haemorrhages and slight or moderate oedema. After 28 hours the dermal effects observed generally consisted of very slight to slight ischemic necrosis, moderate haemorrhages, slight or moderate incrustation and slight oedema.</p> <p>During the course of the following two days ischemic necrosis, haemorrhages and oedema were no longer observed. The application sites generally became crater-shaped with a central sunken area which was moderately or severely encrusted, and a surrounding, raised border of non-necrotic skin showing well-defined erythema.</p> <p>After 7 days this picture had hardly changed, apart from clearance of erythema. The central sunken areas of the application sites generally showed moderate to severe incrustation.</p> <p>At the end of the observation period, after 3 weeks, some signs of healing were observed at the edges of the encrusted skin areas which had been in contact with the test material. In the new skin visible under the crust edges coming off from the treated skin, formation of scar tissue could be observed whereas hair growth was absent. There were no differences between reactions of the intact skin and those of abraded skin. This scar tissue formed already or to be formed is not considered a reversible skin alteration.</p>
Conclusion	On the basis of the results obtained, it can be concluded that lactic acid (88%) is severely irritating to the skin of albino rabbits and that lactic acid (88%) is corrosive to the skin of albino rabbits.
Remark	Other studies have shown that the skin of albino rabbit is not the appropriate animal model when addressing the effects of lactic acid on human skin. This result is therefore not used for the classification.

Title	Acute dermal irritation / corrosion study with PURAC BF S36 and PURAC BF S/30 in albino rabbits
Species:	New Zealand White albino rabbits
Concentration:	buffered lactic acid: BF S36 (38% l.a. + 38% sodium lactate, total 76%

	d.s.). BF S30 (60% l.a. + 20% sodium lactate, total 80% d.s.)
Report number	V 96.677
Year	1996
GLP	Yes
Test substance & purity	buffered lactic acid: BF S36 (38% l.a. + 38% sodium lactate, total 76% d.s.). BF S30 (60% l.a. + 20% sodium lactate, total 80% d.s.)
Exposure	Occlusive
Exposure Time	4 hours.
No. of Animals	3
Experimental design	<p>Four days prior to the start of the study, the hair was removed from the back and the flanks of the animals in a way to avoid abrasions. Each rabbit was treated simultaneously with the same compound.</p> <p>An amount of 0,5 ml of each test substance was distributed over a patch measuring 2.5 x 2.5 cm. The two patches were fixed to the application sites. Subsequently, the entire trunk of the rabbit was wrapped with self-adhesive tape.</p> <p>After a 4 hour exposure period, the test substance and patches were removed and the test sites were cleaned with moistened tissues.</p> <p>After a hour the resulting skin reactions were evaluated by the method of Draize et al (J. Pharmacol. Exp. Ther. 82 (1944) 377-390). Further skin readings were made at approximately 24, 48 and 72 hours after treatment.</p>
Observations	After 1-72 hours after treatment, no signs of skin irritation were observed in any of the three rabbits.
Conclusion	On the basis of the results obtained, it was concluded that, under the conditions of this study, PURAC BF36 and PURAC BF S30 are not irritating for the skin of rabbits after a 4 hour dermal contact period.
Remark	

Title	Primary dermal irritation study in rabbits
Species:	New Zealand White albino rabbits
Concentration:	80%
Report number	410-1355
Year	1983
GLP	Yes
Test substance & purity	L(+) lactic acid 80%, diluted with water
Exposure	Occlusive
Exposure Time	24 hours
No. of Animals	6, 3 male and 3 female
Experimental design	<p>Two test sites were prepared on each side of the spinal column in the thoracic region of each animal by closely clipping the hair.</p> <p>Approximately 24 hours after clipping, 0.5 ml of the test article was applied to each 2.5 cm square surgical gauze patch. 2 Application sites on each animal were abraded with a needle to penetrate the stratum corneum but not the dermis. Four patches containing the test article were then applied to each animal and held in place with gauze wrapping.</p> <p>After 24 hours exposure, the patches were removed. Each test site was gently wiped with gauze sponges moistened with water to remove the remaining test article.</p> <p>The skin conditions of each test site was evaluated for erythema, edema and other lesions at 30 and 60 minutes after test article removal. After 30-60 minutes evaluations all animals were euthanized due to severity of the dermal reactions observed.</p> <p>EPA OPP 81-5</p>
Observations	<ul style="list-style-type: none"> <li>- Severe erythema at 10 of the 12 abraded test sites (on 5 animals) and at 7 of the 12 intact test sites on 4 of the same animals. Moderate to severe erythema was seen at 2 remaining abraded and 5 remaining intact test sites.</li> <li>- Severe edema at 11 of 12 intact test sites and at 11 of 12 abraded test</li> </ul>

	<p>sites. Slight edema at one abraded test site and one intact test site on one animal.</p> <ul style="list-style-type: none"> <li>- Blanching at both abraded sites on each animal and at both intact sites at 5 animals.</li> <li>- Yellow-brown color of skin at all sites on 3 animals, at both abraded sites on one animal and at one intact site and both abraded sites on a fifth animal.</li> <li>- Red exudate at one intact site on one animal.</li> <li>- Skin missing at all sites of one animal, at one intact site and both abraded sites on one animal, and at one intact site or one abraded site on 2 other animals.</li> </ul> <p>No other dermal reactions were observed at the evaluations done at 30 to 60 minutes after test article removal. No abnormal clinical sign were observed and no mortalities occurred prior to sacrifice after 30 – 60 minute evaluations.</p>
Remark	Under the definition of CFR 49, 173.136, the product does not need to be classified

Title	Lactic acid Q88: A skin corrosivity test in guinea pigs
Species:	Guinea pigs, Dunkin Hartly strain
Concentration:	88%
Report number	235943
Year	1986
GLP	Yes
Test substance & purity	L(+) lactic acid 88%, diluted with water
Exposure	Occlusive
Exposure Time	Up to 4 hours
No. of Animals	6
Experimental design	<p>The animals were divided into 2 groups of 3. The hair was clipped from both flanks 24 hours before patch application. Care was taken to avoid abrading of the skin.</p> <p>In group 1, 0.5 ml of the test material was applied (at two sites) under 2 gauze patches each measuring 2.5 x 2.5 cm. The patches were applied to the intact skin and covered with Micropore tape. The whole trunk was then bound with Elastoplast elastic bandage to give a semi-occlusive covering. Group 2 animals were treated in the same way except that the test material was applied under one patch.</p> <p>For group 1 animals the first patch was removed 3 minutes after application and the skin was gently washed with water to remove any residual test material. Reactions were scored. One hour after application the second patch was removed, the skin washed and reactions scored.</p> <p>The patch for group 2 animals was left in place for 4 hours before removal and scoring.</p> <p>Skin reactions were scored for erythema and eschar formation and for oedema formation at patch removal, 1, 24, 48 and 72 hours after patch removal.</p>
Observations	<p>In group 1 ( 3 minutes and 1 hour exposure) no erythema, no eschar formation and no oedema formation was seen.</p> <p>In group 2 ( 4 hours exposure) very slight erythema (barely perceptible) and very slight oedema formation (barely perceptible) was seen at patch removal and after one hour.</p> <p>In group 2 no erythema, no eschar formation and no oedema formation was seen at 24, 48 and 72 hours after patch removal.</p>
Conclusion	It is concluded from the test results that lactic acid 88% is not corrosive to guinea pig skin. It is also noted that irritation after 4 hours exposure was transient and limited to very mild erythema.
Remark	

Title	Acute dermal irritation/corrosion study with lactic acid 88% in pigs
Species:	Healthy male, young pigs from Large white (GY) x Dutch Landrace (NL)
Concentration:	88%
Report number	V 87.405
Year	1987
GLP	Yes
Test substance & purity	L(+) lactic acid 88%, diluted with water
Exposure	Dermal, 0,5 ml test material per application site, occlusive.
Exposure Time	Up to 4 hours
No. of Animals	3
Experimental design	The hair was removed from the animals. The test material was brought into contact with three small separate areas of intact skin. Each of the application sites was covered with occlusive patch measuring 1 inch x 1 inch. The first patch was removed after 3 minutes, the second after 1 hour and the third after 4 hours. Immediately after removing the patches the treated skin areas were treated with lukewarm water, and one hour later the resulting skin reactions were evaluated by the method of Draize et.al. (J. Pharmacol. Exp. Ther. 82 (1944) 377-390). Further readings were made after 1 day and after 2, 3, 7, 14 and 21 days.
Observations	No dermal irritaton responses related to treatment with lactic acid (88%) were observed at the test sites of any animal during the 21-day observation period, etither after a 3 minute contact eperiod, after 60 minutes contact period or after a 4-hour contact period. After one day, superficial wounds were observed at the test site of pig no. 1 treated for 60 minutes and at the test site of pig no. 3 treated for 4 hours. In the same animals, similar injuries were also seen at the non-treated skin. After two days, the same test sites and non-treated skin areas in pigs 1 and 3 showed some slight small crusts. After three days these affects had cleared. These minor injuries were probably caused by shaving along the walls or the floor of the stable.
Conclusion	Under the conditions of this test it is concluded that lactic acid (88%) is not a primairy skin irritant to the pig skin.
<u>Remark</u>	TNO and other experts believe the pig to be a more appropriate and representative animal model than the albino rabbit, when addressing the effects of lactic acid on human skin.

## 6. Eye Irritation

Species	rabbit
Concentration:	20 % L(+) lactic acid
Year	1973
GLP	yes
Test substance & purity	20 % L(+) lactic acid in water
Experimental design	Journal Officiel de la Republique Francaise procedure; eyes were examined after 1 and 24 h and after 2, 3, 4, and 7 days with fluorescent staining.
Observations	In same study also 50% sodium lactate was tested, which is not irritating. Instilled at 20% and 10% provoked significant ocular irritation :Acute Ocular Irritation Index (AOII) was 39.50 resp. 31.17. Only for the 10% dilution these lesions were reversible, 7 days after instillation.
Conclusion	irritating
Remark	<b>From public literature, see references.</b>

Title	Chicken Enucleated eye test with three samples of lactic acid; an alternative to the Draize eye test.
Species:	Male or female chickens (ROSS, spring chickens) were used as eye donor.
Concentration:	
Report number	V96.157
Year	1996
GLP	Yes
Test substance & purity	<ul style="list-style-type: none"> <li>- Powder H60 (60% L(+) lactic acid and 40% calcium S lactate)</li> <li>- HS88, 88% L(+) lactic acid in water.</li> <li>- BF S36 L(+) lactic acid; L(+) lactic acid and sodium S lactae.</li> </ul>
Exposure	Corneal
Exposure Time	10 seconds
No. of Animals	10 eyes
Experimental design	<p>Within 2 hours after the kill the heads were carefully dissected and placed in a superfusion apparatus. The eyes are checked for damages. Corneal thickness was measured. Per test sample three eyes were selected for testing whereas one eye was rinsed with isotonic saline only and served as a control of the experimental conditions.</p> <p>At time t=0 0.03 ml / 0.03 g of the test sample was applied to the eye in such a way that the entire surface of the cornea was bathed with the test material. After a total exposure period of 10 seconds, the corneal surface was rinsed thoroughly with 20 ml of isotonic saline of ambient temperature. The control and the test eyes were examined for changes in corneal thickness (swelling), corneal opacity and fluorescein retention at 0, 30, 75, 120, 180 and 240 minutes after treatment.</p>
Observations	<ul style="list-style-type: none"> <li>- <u>powder H60</u> After treatment the thickness of the cornea of the eyes gradually increased considerably; a maximum mean corneal swelling of 17% was obtained at 240 minutes after treatment. In addition, moderate corneal opacity and moderate fluorescein retention by damaged epithelial cells were observed in the eyes. The categories defined for corneal swelling, corneal opacity and fluorescein retention were II, III and III</li> <li>- <u>PURAC lactic acid HS88</u> After treatment severe to complete corneal opacity was observed in the three test eyes, which hampered the measurement of corneal thickness</li> </ul>

	<p>at the 30, 75 and 120 minutes after treatment. At 180 and 240 minutes after treatment corneal thickness could be measured again and at 240 minutes a maximum corneal swelling of 28% was determined. All three eyes showed severe fluorescein retention by damaged epithelial cells. The categories defined for corneal swelling, corneal opacity and fluorescein retention were III, IV and IV</p> <ul style="list-style-type: none"> <li>- <u>PURAC lactic acid bufered BF S36</u> After treatment only a slight increase in cornela thickness was observed. A maximum corneal swelling of 6% was obtained 75 minutes after treatment. Very slight corneal opacity and slight fluorescein retention by damaged epithelial cells were observed in the test eyes. The categories defined for corneal swelling, corneal opacity and fluorescein retention were II, I and II</li> <li>- <u>Control eye</u> The control eye did not show any unusual effects.</li> </ul>
Conclusion	<ul style="list-style-type: none"> <li>- <u>powder H60</u>  Irritating to the eyes</li> <li>- <u>lactic acid HS88</u>  Severely irritating to the eyes</li> <li>- <u>lactic acid bufered BF S36</u>  Not irritating to the eye.</li> </ul>
Remark	The measurement of the cornela swelling in this test guarantees a highly objective parameter, which enables the investigator to discriminate the damaging effects of the test material very precisely, this in contrast to the conventional rabbit test which uses subjetive gross measurement only.

Species:	other
Concentration:	85 %
Experimental design	<u>Test condition:</u> in vitro using the Eytex Assay (Avon Products, Inc, 1995). Most of the formulations were tested undiluted. pH of formulations varies from 7.52 to 2.02.
Observations	Only formulation with pH 2.02 (face cream with 11.8% lactic acid 85%) was moderate severe irritant. The formulations with pH $\geq$ 5.3 were minimal irritant.
Conclusion	irritating
Remark	<b>From public literature, see references.</b>



## 7. Sensitisation

Title	Dermal sensitization study in guinea pigs (Buehler Test)
Species:	Female, Hartley guinea pigs
Concentration:	L(+) lactic acid 80%, diluted with water
Report number	480-2750
Year	1986
GLP	Yes
Test substance & purity	L(+) lactic acid 80%, diluted with water
Exposure	Dermal
Exposure Time	Two days for the range finding and 35 days for the main study.
No. of Animals	22 (2 for range finding, 10 for study test group and 10 for study naïve group)
Experimental design	The hair on the back or left flank (induction) and/or flank (range finding and challenge) of each animal was closely clipped. A 0.5 m sample of the test article and 0.5 ml samples of 30, 10 or 5 % suspensions of the test article in dionized water were placed on separate patches and applied to a range finding animal. The 100% test article was selected for induction.
Observations	<p>Preliminary range-finding trials revealed very slight erythema at the 100% concentration of the test article. No other dermal reactions were noted for the other concentrations (3, 10 and 30%). Therefor the 100% concentration of the test article was utilized in the main study testing for contact dermal sensitization potential.</p> <p>No mortalities occurred and all animals gained body weight. The test article (100%) produced very slight erythema at 3 sites and very slight edema at 1 site after the first induction. Erythema grades increased in severity after the second induction application. Due to the increase of severity of the reactions, the concentration of the test article was reduced to 30% and the induction site was changed to the left flank. Very slight erythema was noted after the fifth induction application. Grades ranging from very slight to severe erythema were noted from the seventh to the ninth induction applications.</p> <p>After the challenge application the test article (100%) produced grade 4 erythema in up to six test animals and in up to eight control animals. These reactions were considered irritation reactions, not sensitization reactions. The reactions seen in the control animals at the challenge were similar to the reactions seen for the test group animals and the test article was not considered to be a contact dermal sensitizer.</p>
Conclusion	The reactions seen (very slight to moderate erythema, very slight to moderate oedema) were considered to be irritant reactions, not sensitive reactions. The test article was not considered to be a dermal sensitizer.
Remark	

## 8. Repeated Dose Toxicity

Type	Sub-chronic
Species:	rat
Concentration:	4 ml lactic acid 10% on 20 g of meal
Exposure Time	Exposure period: 90 days
Experimental design	Route of administration: gavage Control Group: yes, concurrent no treatment Frequency of treatment: every day

Observations	No differences in appearance, gross observations at necropsy, or organ weights were observed between the test and control animals. Changes in blood carbon dioxide were slight.
Conclusion	
Remark	<b>From public literature, see references.</b>

Type	Sub-chronic
Species:	female Sprague-Dawley rat
Test substance & purity	formulation (face cream containing 0.25% of lactic acid 85%)
Exposure	dermal
Exposure Time	13 weeks treatment daily, 5 days/week Control Group: yes, concurrent no treatment
Experimental design	Doses: 886 mg/kg bw
Observations	No significant gross observations, with the exception of minimal skin irritation. Absolute brain weight and kidney-to-body weight ratios were increased for test animals. No lesions were observed at necropsy or at microscopic examination.
Conclusion	The formulation is safe in terms of cumulative toxicity. Based upon the exaggerated dose level used in this study for skin care products, dermal application is not likely to produce adverse effects under conditions of consumer use. LOAEL: 886 mg/kg
Remark	<b>From public literature, see references.</b>

Type	Sub-chronic
Species:	Fischer 344 male/female rat
Concentration:	experiment I: calcium lactate dissolved in drinking water (up to 5%). experiment II: up to 30% calcium lactate in diet.
Test substance & purity	Calcium lactate as a salt of lactic acid.
Exposure	Oral; 5, 2.5, 1.25, 0.6, 0.3 %
Exposure Time	Exposure period: 13 weeks
Experimental design	Control Group: yes, concurrent no treatment
Observations	a <10 % decrease in body weight gain. all animals survived. some haematological and biochemical parameters changed, but no severe lesions were found in microscopic examination in the experiment with calcium lactate mixed in the diet, the amount of calcium in the urine was significantly increased. Nephrocalcinosis and degeneration in kidneys observed. Indications that Nephrocalcinosis was dependent on the low Ca/Phosphorus ratio of the synthetic diet.
Remark	Lactic acid tested as its Calcium salt. From this study the lactate part is relevant, should be separated from effects of the soluble Calcium intake. <b>From public literature, see references.</b>

Type:	Sub-chronic
Species:	Syrian hamster male/female
Exposure	Group 1 (control): Diet 1, contains 20% sucrose as carcinogenic diet; pure water to drink. Group 2: diet 1, mixed with 0.057 ml lactic acid 80%; pure water to drink. Group 3: same diet 1, but water containing 0.050% v/v lactic acid
Exposure Time	14 weeks
Experimental design	daily ad libitum; animals of groups 2 & 3 ingest same amount lactic acid. Post exposure period: sacrificed and autopsy; also oral cavity (caries incidence) Control Group: yes, concurrent no treatment

Observations	three groups same growth and health. No significant differences were found in the incidence or extent of carious lesions among the three groups.
Conclusion	dietary lactic acid did not play any important role in development or progress of dental caries.
<u>Remark</u>	pH of diet 1 is 5.55, of diet 2 is 5.12. pH of pure water is 6.8 and of water + lactic acid is 3.1 . <b>From public literature, see references.</b>

## 9. Genetic Toxicity 'in Vitro'

Type:	Ames test
Species:	
Concentration:	0.5, 1.0, and 2.0 microliter lactic acid/plate
Report number	
Year	
GLP	
Test substance & purity	
Exposure	
Exposure Time	
No. of Animals	
Experimental design	Salmonella/microsome test (Ames test) and chromosomal aberration test in vitro reverse mutation assays, and Chinese hamster fibroblast cell line S. typhimurium strains TA97, TA98, TA100, TA104 Metabolic activation: with and without
Observations	negative
Conclusion	
Remark	<b>From public literature, see references.</b>

Type:	Chinese hamster ovary K1 cells, chromosomal aberration tests, and the pH relationship of the medium and clastogenic activity was examined.
Concentration:	8-35 mM
Experimental design	Cells were maintained in Ham's F12 medium, supplemented with 10% foetal calf serum. Cytotoxic Concentration: 14-35 mM, when pH was $\leq 5.8$ Metabolic activation: with and without
Observations	When the culture medium was first acidified by the lactic acid dose and then neutralised to pH 6.4 or when medium is containing 30 mM HEPES as buffer, lactic acid was non-clastogenic. Pseudo-positive reactions are seen as a result of non-physiological low pH.
Conclusion	lactic acid was non-clastogenic
Remark	<b>From public literature, see references.</b>

Type	review on several mutagenicity studies with lactic acid and some lactates.
Concentration:	various
Experimental design	various Metabolic activation: with and without
Conclusion	negative
Remark	the result of 11 studies is reviewed <b>From public literature, see references.</b>

## 10. Genetic Toxicity ' in Vivo '

Due to the natural nature of L(+) lactic acid and the relative low contribution of "outside L(+) lactic acid" to the human metabolism, in vivo genotoxicity studies will not be required.

### ▪ **Carcinogenicity**

Title	
Species:	Fischer 344 male/female rat
Concentration:	2.5 or 5 % Calcium lactate in the drinking water. Mean total Calcium lactate intake for males was 329.4 g, resp. 625.4 g; for females 237.7 g, resp. 412.1 g.
GLP	yes
Test substance & purity	The Calcium salt of lactic acid was tested
Exposure	drinking water daily, ad lib
Exposure Time	2 years
Experimental design	Control Group: yes, concurrent no treatment
Observations	Autopsy on rats that died during study and those killed at the end. Examination macro-and microscopically for presence of non-neoplastic and neoplastic lesions
Conclusion	Negative
Remark	Lactic acid tested as its Calcium salt. From this study the lactate intake is relevant, should be separated from the Calcium effects of a soluble Calcium salt. <b>From public literature, see references.</b>

Species:	female rabbit
Concentration:	0.1-0.2 g/kg bw (5 months), and 0.1-0.7 g/kg bw (13 months)
Exposure	drinking water twice daily
Exposure Time	5 or 13 months
Observations	No tumors were reported after 5 or 16 months. Further details not provided.
Conclusion	negative
Remark	<b>From public literature, see references.</b>

## 11. Toxicity to Fertility

The nature of the compound (part of human metabolism) does make toxicity studies to fertility not necessary

## 12. Developmental Toxicity/Teratogenicity

Species:	female CD-1 mouse
Exposure Time	gestational days 6-15 Duration of test: 10 days
Experimental design	Administration: gavage Frequency of treatment: daily Doses: 570 mg/kg bw/day Control Group:yes, concurrent no treatment
Observations	Lactic acid was neither maternotoxic nor embryofetotoxic when given orally to mice at 570 mg/kg bw/day on gestation days 6-15.
Conclusion	NOAEL Maternal Toxicity: $\geq 570$ mg/kg bw NOAEL Teratogenicity: $\geq 570$ ml/kg bw
Remark	<b>From public literature, see references.</b>

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